



Lawrence Berkeley Laboratory

UNIVERSITY OF CALIFORNIA

CHEMICAL BIODYNAMICS DIVISION

Submitted to Biochemistry

THE PSORALEN-DNA PHOTOREACTION. CHARACTERIZATION OF
THE MONOADDITION PRODUCTS FROM 8-METHOXYPsorALEN AND
4,5',8-TRIMETHYLPSORALEN

David Kanne, Kenneth Straub, Henry Rapoport, and
John E. Hearst

July 1981

RECEIVED
LAWRENCE
BERKELEY LABORATORY

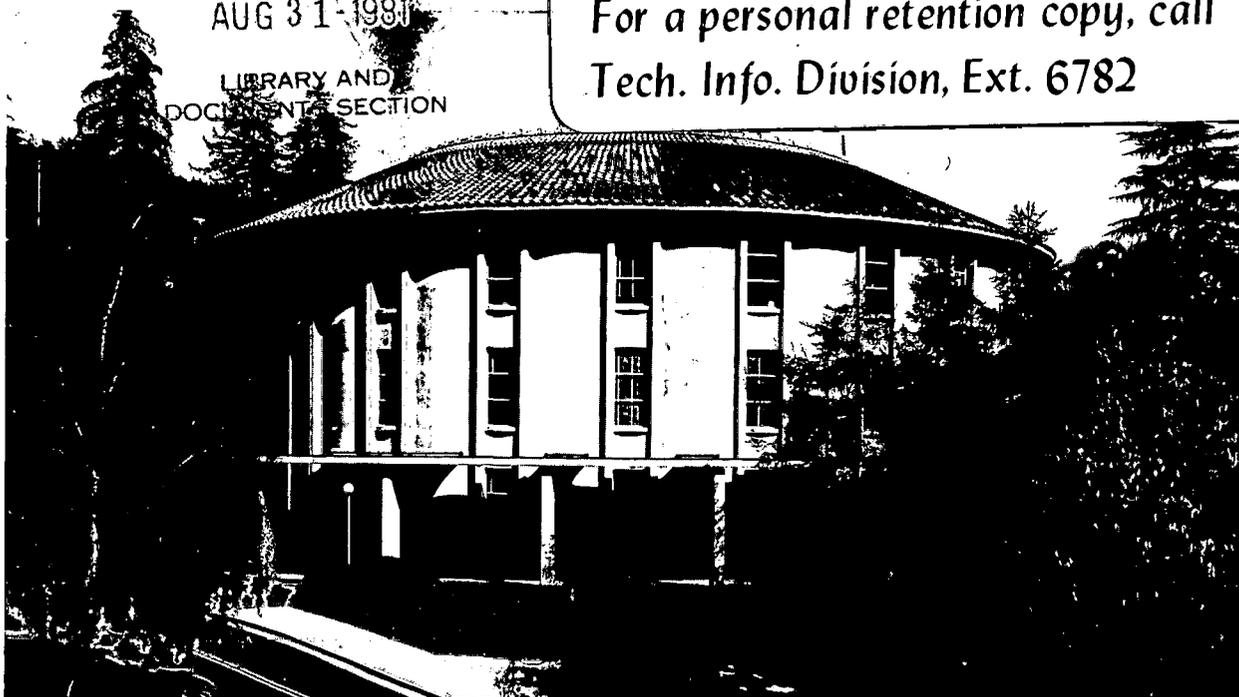
AUG 31 1981

LIBRARY AND
DOCUMENTS SECTION

TWO-WEEK LOAN COPY

*This is a Library Circulating Copy
which may be borrowed for two weeks.*

*For a personal retention copy, call
Tech. Info. Division, Ext. 6782*



LBL-13083
2

DISCLAIMER

This document was prepared as an account of work sponsored by the United States Government. While this document is believed to contain correct information, neither the United States Government nor any agency thereof, nor the Regents of the University of California, nor any of their employees, makes any warranty, express or implied, or assumes any legal responsibility for the accuracy, completeness, or usefulness of any information, apparatus, product, or process disclosed, or represents that its use would not infringe privately owned rights. Reference herein to any specific commercial product, process, or service by its trade name, trademark, manufacturer, or otherwise, does not necessarily constitute or imply its endorsement, recommendation, or favoring by the United States Government or any agency thereof, or the Regents of the University of California. The views and opinions of authors expressed herein do not necessarily state or reflect those of the United States Government or any agency thereof or the Regents of the University of California.

The Psoralen-DNA Photoreaction. Characterization of the Monoaddition
Products from 8-Methoxypsoralen and 4,5',8-Trimethylpsoralen[†]

[†]From the Department of Chemistry and Lawrence Berkeley Laboratory,
University of California, Berkeley, California 94720

This research was supported in part by the Office of Environmental Research and Development, Biomedical and Environmental Research Division of the DOE under Contract No. W-7405-ENG-48, and the National Institute of General Medical Sciences (Grant GM-25151).

David Kanne, Kenneth Straub, Henry Rapoport* and John E. Hearst

Running Title

Psoralen-DNA Photoreaction Products

This manuscript was printed from originals provided by the authors.

Footnotes

¹This deoxythymidine adduct derives primarily from photoaddition to a deoxythymidine residue. However, should any adduct formation have occurred with the small amount of 5-methylcytosine present (Ehrlich & Yang, 1981) it would have been hydrolytically deaminated (Freeman et al, 1965) and appear as a deoxythymidine adduct.

²Those adducts obtained by acid hydrolysis are indicated by the addition of H⁺ to the fraction number.

ABSTRACT: The isolation and structural characterization is described of the major monoaddition products formed in the photoreaction of two naturally occurring psoralens, 8-methoxypsoralen and 4,5',8-trimethylpsoralen, with high molecular weight, double-stranded DNA. Hydrolysis of the psoralen-modified DNA and subsequent chromatography resulted in the isolation of four modified nucleosides from each psoralen. Structural characterization was accomplished by mass spectrometry and $^1\text{H-NMR}$ analysis. The major products, accounting for 50-60% of the covalently bound psoralen, are two diastereomeric deoxythymidine adducts formed by cycloaddition between the 5,6-double bond of the pyrimidine and the 4',5'-(furan) double bond of the psoralen. A minor product, less than 2% of the covalently bound psoralen, is a furan-side adduct to deoxyuridine, derived from an initially formed deoxycytidine adduct by hydrolytic deamination. A fourth product is a deoxythymidine adduct where cycloaddition has taken place between the 5,6-double bond of the pyrimidine and the 3,4-(pyrone) double bond of the psoralen. This pyrone-side adduct accounts for 18% of the covalently bound 8-methoxypsoralen, but for less than 3% of the covalently bound 4,5',8-trimethylpsoralen. All of the isolated adducts have cis-syn stereochemistry. The stereochemistry and product distribution of the adducts is determined in part by the constraints imposed by the DNA helix on the geometry of the non-covalent intercalation complex formed by psoralen and DNA prior to irradiation.

Psoralens or furocoumarins are a class of compounds found in a wide variety of plants and fungi and have been used since ancient times as dermal photosensitizing agents for the treatment of various skin pigmentation disorders (Scott et al, 1976). Their photosensitizing activity is dependent upon subsequent exposure to near ultraviolet light (300-380 nm), and is now known to involve photochemical addition to the DNA of the target tissue (Song & Tapley, 1979). This addition to cellular DNA involves both monoaddition and interstrand cross-linking, as evidenced by denaturation-renaturation kinetics and electron microscopy (Cole, 1970; Cole, 1971; Hanson et al, 1976). The ability of psoralens to crosslink double-stranded DNA has attracted interest in their use as probes of nucleic acid structure and function (Wieseahn et al, 1977; Shen & Hearst, 1977), and they are also under investigation as therapeutic agents for the treatment of psoriasis (Wolff et al, 1977). In addition, their ability to interact efficiently with DNA has drawn attention to their potential as naturally occurring mutagens and carcinogens. The mutagenic activity of compounds such as 8-methoxypsoralen (xanthotoxin) and 5-methoxypsoralen (bergapten) is well established, and reports have appeared documenting the tumorigenic activity of psoralens in a number of species, including man (Ashwood-Smith et al, 1980; Stern et al, 1979; Epstein, 1979).

The photoreaction between psoralens and DNA can be divided into at least three distinct steps: (1) formation of a noncovalent complex with DNA via intercalation of the psoralen between adjacent base pairs, (2) photoreaction between the psoralen and a pyrimidine base to yield a monoadduct, (3) absorption of a second photon to yield an interstrand cross link. The interstrand crosslinks are believed to be largely responsible for the photosensitizing effects of psoralen treatment, although some activity is apparently associated with the monoadducts (Harten et al, 1976). The chemistry of the photoreaction has been studied intensively, but detailed information about the structures of the nucleic acid-psoralen adducts has been notably lacking. The study reported here is concerned with the isolation and identification of the monoaddition products formed from the photobinding to DNA of two of the most widely used naturally occurring psoralens, 8-methoxy-psoralen (xanthotoxin or 8-MOP, 1) and 4,5',8-trimethylpsoralen (trioxsalen or TMP, 2).

We have isolated and characterized four monoaddition products derived both from the reaction of 1 or 2 with native, double-stranded DNA. Two of the products are diastereomeric psoralen-deoxythymidine adducts, formed by cycloaddition between the 5,6-double bond of the pyrimidine and the 4',5'-(furan) double bond of the psoralen. A third and minor product (less than 2% of the isolated monoadducts) is a

Struc.
1,2
here

furan-side psoralen-deoxyuridine adduct, derived from an initially formed psoralen-deoxycytidine adduct by hydrolytic deamination. A fourth monoadduct recovered from both 8-MOP- and TMP-modified DNA has been characterized as an adduct resulting from cycloaddition of deoxythymidine residues to the 3,4-(pyrone) double bond of the psoralen. In the case of 8-MOP, this pyrone-side adduct accounts for 18% of the total covalently bound psoralen, while in the case of TMP only 2-3% of the bound psoralen is present as pyrone side adducts.

Experimental Section

Materials [^3H]-8-Methoxypsoralen and [^3H]-4,5',8-trimethylpsoralen were synthesized as described (Isaacs et al, 1981). Hydrolytic enzymes and calf thymus DNA were obtained from Sigma (St. Louis, MO).

Photobinding. A 4.05 mg sample of [^3H]-8-MOP (sp. act. 0.1-0.5 Ci mmol^{-1}) in 1.5 mL of ethanol was added to a solution of 43.2 mg of calf thymus DNA in 60 mL of tris buffer (10 mM tris, pH 7.2). This solution was irradiated for 2.5 hr at 20°. Photobinding of [^3H]TMP (0.1-0.5 Ci mmol^{-1}) to DNA was carried out by adding five 300 μL aliquots of a stock solution (1.25 mg TMP in 1.5 mL ethanol) to the DNA solution (7.2 mg of DNA at a concentration of 0.12 mg mL^{-1} in tris buffer). The solution was irradiated for ten minutes after each of the psoralen

additions. Two 400W GE mercury vapor lamps were used for the irradiation, and cooling of the DNA solution was achieved by circulation of a solution of cobaltous nitrate (40% w/w) through an outer jacket; this solution also served as a 365 nm transmission filter. After the appropriate irradiation time, the psoralen-DNA solution was extracted with four volumes of chloroform to remove unreacted psoralen and its photodegradation products. The remaining solution was made 0.2M in sodium chloride, and diluted with three volumes of cold ethanol to precipitate the DNA which was collected by centrifugation, dissolved in 0.2M sodium chloride, and reprecipitated by addition of ethanol. The isolated DNA pellet was then dried under vacuum and redissolved in hydrolysis buffer (15 mM sodium acetate, pH 5.00). Alternatively, the DNA was subjected to acid hydrolysis, as described below.

Adduct Isolation. The psoralen-modified DNA was first hydrolyzed by the addition of 80 units of DNAase-II (E.C.3.1.4.5) per mg of DNA. After 12 hr, the solution was adjusted to pH 7, and 0.2 units of phosphodiesterase-II (E.C.3.1.4.18) per mg of DNA was added. After an additional 12 hr, the pH was adjusted to 8.0, and 0.2 units of alkaline phosphatase (E.C.3.1.3.1) per mg of DNA was added. After a total hydrolysis time of 24-36 hr, the mixture was lyophilized and redissolved

in water. This mixture was then analyzed by HPLC. HPLC was carried out on reverse-phase Ultrasphere ODS columns (either 4.6 or 10 mm x 25 cm; Altex-Beckmann, Berkeley, CA). Water-methanol was used as the eluting solvent, at a flow rate of 1 or 4 mL min⁻¹. In cases where weak ion exchange effects led to peak broadening, a buffered aqueous phase was used (10 mM KH₂PO₄, pH 2). The column effluent was monitored for absorbance at 254 nm, and assayed for the presence of ³H by scintillation counting.

Mass Spectrometry. Adduct fractions isolated by HPLC were taken to dryness, and either analyzed directly by field desorption mass spectrometry or derivatized then analyzed by high resolution electron impact mass spectrometry. Samples were converted to pertrimethylsilyl ethers by heating 0.5-2.0 µg of adduct in 50 µL of pyridine-BSTFA (N,O-bis-trimethylsilyltrifluoroacetamide, 1:4 v/v) at 60° for 40 min under nitrogen.

Mass spectra were recorded on a modified Kratos/AEI MS902, interfaced to a Sigma 7-LOGOS II data system (Burlingame et al, 1974). Field desorption spectra were obtained using conventional benzonitrile-activated emitters (Beckey, 1977). Samples analyzed by electron impact mass spectrometry were admitted to the ion source via the direct insertion probe (source temperature 250°). Electron-impact mass spectra were recorded at a dynamic resolution of 10000 (scan rate 8 or 16 sec/

decade), and elemental compositions were then generated for a given error tolerance for all fragment ions in the mass spectrum.

$^1\text{H-NMR}$. $^1\text{H-NMR}$ spectra were recorded on a Nicolet Technologies NT-360 spectrometer, or on a Bruker HXS-360 spectrometer. Typically, 256-2048 transients were accumulated at a spectral width of 1500 Hz (16K data points). For nuclear Overhauser effect experiments, the decoupling field was gated off during data acquisition, and a delay time of 3-3.5 seconds was inserted between the end of data acquisition and the beginning of the following pulse. With samples where the size of the residual solvent signal (H₂O) was a problem, a weak saturating pulse was applied prior to the main RF pulse in order to selectively suppress the solvent signal. Spectra used for detecting NOE enhancements were first zero-filled to give 32K data points, and then subtracted from the reference spectrum, usually a spectrum with the decoupling field set symmetrically on the opposite side of the carrier frequency. All assignments were made with the aid of extensive homo-nuclear decoupling experiments.

Spectra were recorded in either 99.996% D₂O or 99.96% d₆-DMSO. Samples were prepared by dissolution in 99.96% D₂O, and then evaporation to dryness under reduced pressure. This was repeated a total of four times. All final loading steps were carried out under a dry nitrogen atmosphere. Typical sample concentrations ranged from

50-500 μM . Chemical shifts are referenced to TMS ($\delta_{\text{HDO}}=4.75$ ppm).

Acid Hydrolysis. DNA samples were dissolved in 0.4N HCl (1 mg mL^{-1}) and heated to 65° for 4 hr. The hydrolyzed mixture was reduced to a minimal volume, and applied to a C-18 Sep Pak cartridge (Waters, Milford MA). Elution was carried out with water and then with methanol, and the methanol fraction was analyzed by HPLC.

Results: HPLC and UV

The results of the photobinding of 8-MOP and TMP to double-stranded, high molecular weight DNA are shown in Table I. The modification levels of 65 and 110 psoralen residues per 1000 base pairs, respectively, are similar to the levels reported in the literature by others (Rodighiero et al, 1970). The psoralen-modified DNA's were then hydrolyzed enzymatically and the resulting hydrolysis mixture contained the psoralen-modified nucleosides, along with a much larger amount of unmodified nucleosides. Isolation of the individual adducts from this complex sample matrix was accomplished by high performance liquid chromatography (HPLC).

Figures 1a and 1b show the HPLC elution profiles obtained from enzymatically hydrolyzed [^3H] 8-MOP-DNA and [^3H]TMP-DNA, respectively. The peaks eluting between 5 and 20 minutes do not contain ^3H , and correspond to the unmodified nucleosides (deoxythymidine, deoxycytidine,

deoxyguanosine, and deoxyadenosine). The major ^3H -containing peaks in the case of 8-MOP-modified DNA are F32, F36, F37, F38, and F41 (Figure 1a and Table I). Unreacted 8-MOP elutes at 51 minutes in this chromatographic system. These five fractions together account for greater than 90% of the covalently bound ^3H . Similarly, TMP-modified DNA yielded four major ^3H -containing fractions; these are referred to as F35, F47, F49, and F52. Unreacted TMP has a retention time of 62 minutes under these conditions (Figure 1b and Table I).

The fractions with elution times of between 36 and 55 minutes in this chromatographic system contain psoralen-nucleoside monoadducts, and it is the characterization of these monoadducts that is the subject of this report. With both 8-MOP and TMP, substantial amounts of interstrand cross-links are formed during these irradiations. These nucleoside-psoralen-nucleoside adducts have elution times of between 28 and 35 minutes, and the structures of these diadducts are currently under investigation.

The absorption spectra of the isolated fractions from 8-MOP or TMP-modified DNA were found to be of two distinct types. Adducts F36, F37, and F38 from 8-MOP and TMP adducts F47 and F49 had very similar absorption spectra with a characteristic maxima at 328 nm (Figure 2). This spectrum is that of a 4',5'-dihydropsoresalen chromophore, and indicates that reaction has occurred at the 4',5'-double bond. The

Table I
here

Fig. 2
here

remaining monoadducts (8-MOP-F41 and TMP-F52) had absorption spectra with the 328 nm band absent (Figure 2). This spectrum is that of a 3,4-dihydropsoresalen derivative, and indicates that reaction has taken place at the 3,4-(pyrone) double bond of the psoralen.

Results: MS

The four 8-MOP nucleoside HPLC fractions (F36, F37, F38, F41) and the three TMP-nucleoside fractions (F47, F49, F52) were analyzed by mass spectrometry. The high polarity and thermal ability of nucleosides generally rule out the direct use of conventional electron impact techniques. Molecular weight information was therefore obtained by the use of field desorption mass spectrometry (FDMS). The adducts were then converted to N,O-pertrimethyl silyl ethers and analyzed by high resolution electron impact mass spectrometry. The resulting fragmentation pattern and elemental composition information, together with the molecular weight information from FDMS, allowed the assignment of overall structures to the adducts.

Analysis by FDMS gave identical results for 8-MOP fractions F38 and F37, indicating a molecular weight of 458 (Table II). 8-MOP-F36 gave results consistent with a molecular weight of 444. Analysis of the TMP adducts indicated a molecular weight of 470 for TMP-F49 and 456 for TMP-F47 (Table III). There was insufficient material present in the 8-MOP-F41 or TMP-F52 fractions to obtain an FD mass spectrum.

Fig. 3
here

A region of the high resolution electron impact mass spectrum of the TMS derivative of 8-MOP-F38 is shown in Figure 3; 8-MOP-F37 gave an identical high resolution spectrum, indicating that the two compounds are probably diastereomers. The highest observed mass occurs at m/z 587 ($C_{21}H_{17}O_9N_2TMS_2$), and corresponds to the loss of a methyl radical from the molecular ion. Additional fragmentation reactions include loss of the elements of CH_2O (m/z 557) and C_2H_4O (m/z 543) from m/z 587. Fragmentations which are characteristic of TMS derivatives of nucleosides include the loss of HOTMS (m/z 497), and cleavage of the C-N glycosidic bond with hydrogen rearrangement to yield the psoralen-base fragment at m/z 343 ($C_{17}H_{15}O_6N_2$). The region below m/z 220 is dominated by a set of fragment ions identical to those observed in the mass spectrum of 8-MOP. The base peak in the spectrum occurs at m/z 216. The nitrogen content (N_2) and overall fragmentation pattern of these two adducts are consistent with a 1/1 8-MOP-deoxythymidine adduct. The 2' and 5' hydroxyls of the deoxyribose have been converted to TMS ethers, while the pyrimidine and lactone moieties remain intact under the derivatization conditions used. The molecular weight obtained by FDMS (458) is consistent with this assignment.

While the mass spectral results alone do not directly identify the site of reaction in the psoralen, together with the uv absorption spectra of the adducts (indicating a 4',5'-dihydropsoalen-type chromophore),

the data strongly suggest a 1/1 8-MOP-deoxythymidine adduct formed via cycloaddition to the 4',5'-double bond of the psoralen.

The HRMS results obtained on 8-MOP-F36 were very similar to those obtained on F37 and F38, except that all of the masses above m/z 300 are shifted to lower mass by 14 units. The elemental compositions of the various fragment ions indicate that the nucleoside base moiety contains $C_4H_3N_2O_2$, ruling out the presence of either thymine or cytosine. The molecular weight and elemental composition data are consistent with the presence of uracil, however. 8-MOP-F36 is therefore a 1/1 8-MOP-deoxyuridine adduct, apparently formed by hydrolytic deamination of an initially formed deoxycytidine-psoralen adduct. This reaction has been observed in other 5,6-dihydrouracil-type compounds (Freeman et al, 1965). The absorption spectrum of F36 indicates that the cycloaddition has occurred at the 4',5'-double bond of the psoralen.

The results obtained on the TMP-DNA adducts F49 and F47 are shown in Table III. In both cases the mass spectrum is dominated by fragment ions from the TMP moiety (base peak = m/z 228), and the region below m/z 300 is very similar to the spectrum of TMP itself. The highest observed mass occurs at m/z 598 ($C_{23}H_{21}O_8N_2TMS_2$), corresponding to loss of a methyl radical from the molecular ion. The fragmentation pattern is similar to that observed with the 8-MOP adducts, and includes ions derived from the loss of the elements of CH_2O (m/z 569), C_2H_4O (m/z 555),

HOTMS (m/z 509), and cleavage of the glycosidic bond (m/z 355).

Together with the FDMS results, this indicates that F49 contains a 1/1 TMP deoxythymidine adduct,¹ and that F47 contains a 1:1 TMP-deoxyuridine adduct. The absorption spectra of these adducts are consistent with cycloaddition having occurred at the 4',5'-double bond of the psoralen.

The remaining products, 8-MOP-F41 and TMP-F52, gave high resolution mass spectra (as TMS ethers) markedly different from that of the other adducts. For TMP-F52 the base peak in the spectrum occurred at m/z 390 ($C_{14}H_{12}O_4TMS_2$) rather than m/z 228. This suggests that hydrolysis and derivatization of the lactone moiety has occurred, resulting in a di-TMS derivative of TMP. An additional fragment at m/z 386 ($C_{10}H_{12}O_5N_2TMS_2$) indicates the presence of deoxythymidine. These results are consistent with a 1/1 TMP-deoxythymidine adduct in which the lactone moiety of the psoralen has undergone hydrolysis to the hydroxy-acid and subsequent conversion to a TMS ether/TMS ester. Facile ring opening in the lactone is characteristic of a 3,4-dihydropsooralen, and the absorption spectrum of this adduct is consistent with the absence of a 3,4-double bond. TMP-F52 is therefore most probably a 1/1 deoxythymidine-TMP adduct, where cycloaddition has occurred on the pyrone (3,4-double bond) side of the psoralen. Extensive fragmentation apparently takes place with the tetra-TMS derivative of this adduct, so that few ions directly related to the molecular ion are observed. The

highest observed mass occurs at m/z 516 ($C_{19}H_{18}O_6N_2TMS_2$) and corresponds to cleavage of the glycosidic bond. There was insufficient material available for FDMS or 1H -NMR; further work to confirm the identity of this adduct is in progress.

The mass spectrum of the TMS derivative of 8-MOP-F41 was similar to that of TMP-F52 in that ions diagnostic for the presence of a di-TMS derivative of the psoralen occur at m/z 378 [$C_{12}H_8O_5(TMS)_2$] and 363 [$C_{11}H_5O_5(TMS)_2$]. Additional ions indicating the presence of deoxythymidine occur at m/z 386 [$C_{10}H_{12}N_2O_5(TMS)_2$] and 126 ($C_5H_6N_2O_2$). These data, together with the ultraviolet absorption spectrum, are consistent with the presence of a 1/1 8-MOP/deoxythymidine adduct where cycloaddition has occurred at the 3,4-(pyrone) double bond of the psoralen.

The psoralen-DNA cycloaddition which occurs at the 5,6-double bond of a pyrimidine base greatly facilitates the susceptibility towards acid hydrolysis of the C-N glycosidic bond of the modified nucleoside. Acid hydrolysis of the psoralen-modified DNA was therefore used to isolate and characterize the aglycone derivatives of the psoralen-modified residues.² Hydrolysis of 8-MOP DNA in mild acid (0.4N HCl, 60°, 2 hr) and subsequent chromatography yielded three monoadducts, referred to as 8-MOP-F39H⁺, 8-MOP-F40H⁺, and 8-MOP-F42H⁺, in approximately a 5/80/15 ratio. The absorption spectrum of the major product,

8-MOP-F40H⁺, indicated the presence of a 4',5'-dihydropsoresalen chromophore and is different from that of the other two fractions. 8-MOP-F39H and 8-MOP-F42H⁺ were found to interconvert at room temperature and neutral pH; at elevated temperatures in mild acid (50°, pH 2), the formation of F42H⁺ is favored over F39H⁺. F42H⁺ has an absorption spectrum similar to that of TMP-F52, indicating that reaction has occurred at the 3,4(pyrone) double bond of the psoralen. This interconversion between F39H⁺ and F42H⁺ is most likely due to hydrolysis of the lactone to a δ -hydroxy acid. Both MS and NMR results obtained on F42H⁺ are consistent with the presence of a lactone moiety.

These aglycon adducts were sufficiently volatile to allow direct HRMS analysis without the need for conversion to TMS derivatives. The data in Table V indicate that a small molecular ion (<5% of base peak) is observed for 8-MOPF40H⁺ (m/z 342, C₁₇H₁₄N₂O₅). The spectra are otherwise very similar to that observed for free 8-MOP, with the exception of ions at m/z 126 (C₅H₆N₂O₂) and 83 (C₄H₅NO). These two ions are diagnostic for the presence of thymine (McCloskey, 1974). The fragmentation patterns are therefore consistent with furan-side or pyrone-side 1/1 8-MOP-thymine adducts, and the FD mass spectra confirmed this assignment (Table IV).

Acid hydrolysis of TMP-modified DNA resulted in a single major monoadduct, TMP-F50H⁺. The mass spectrum of this adduct gave a molecular

Tables
IV, V
here

ion at m/z 354 ($C_{19}H_{18}N_2O_5$), along with ions indicative of the presence of thymine at m/z 126 ($C_5H_6N_2O_2$) and 83 (C_4H_5NO). The FD mass spectrum of this adduct was consistent with its assignment as a 1/1 TMP-thymine adduct. The UV absorption spectrum was identical to that of a 4',5'-dihydro-TMP derivative.

The relationship of the major acid hydrolysis products to the nucleoside-psoralen adducts obtained by enzyme hydrolysis was determined by subjecting the pure nucleoside adducts to identical hydrolysis conditions (0.4N HCl, 60°, 2 hr). Subsequent chromatography revealed that 8-MOP F40H⁺ is derived from both 8-MOP-F37 and 8-MOP-F38, while 8-MOP-42H⁺ is derived from 8-MOP-F41. The major TMP-DNA acid hydrolysis product, TMP-F40H⁺, is derived from TMP-F49. Various minor products were observed in the monoadduct fraction of the acid hydrolysates of 8-MOP or TMP-modified DNA, but were not present in amounts that allowed for characterization. It is possible that the aglycone derivatives of 8-MOP-F36, TMP-F47, and TMP-F52 were present in these uncharacteristic fractions.

Results: ¹H-NMR

8-MOP DNA Adducts. The 360 MHz ¹H-NMR spectra of F37 and F38 are shown in Figures 4 and 6. Both adducts give similar spectra, with various small shift differences. The 4' and 5' protons of the 8-MOP moiety and

the 5-CH₃ and 6-H of the thymidine residue show upfield shifts, indicating that the 4',5'-furan double bond of the psoralen and the 5,6-double bond of the pyrimidine have undergone reaction. These data are consistent with a cyclobutane structure derived from a 2+2 cycloaddition of a thymidine residue to the 4',5'-double bond of 8-MOP.

The regioselectivity of these furan-side monoadducts can be assigned by homonuclear decoupling analysis of the protons of the cyclobutyl system. The downfield 5'-H of the psoralen is coupled equally to both the 4'-H of the psoralen and the 6-H thymidine proton, resulting in an apparent triplet at 5.55 ($J_{4',5'} = J_{5',6} = 5.5$ Hz). This pattern is consistent only with a syn orientation about the cyclobutane ring; in the anti orientation the H-5' psoralen proton would undergo vicinal coupling only to the psoralen 4'-H. There is a small cross-coupling (4J) between the protons positioned diagonally to each other on the cyclobutane ring (4'-H of the psoralen and 6-H of the thymidine). This cross-coupling is readily observed for the 6-H thymidine resonance (a doublet of doublets), but in the case of the psoralen 4'-H resonance it is masked by 4J coupling between 4'-H and 5-H of the psoralen. Irradiation of the 5-H resonance yields a decoupled spectrum in which the mutual 1.5 Hz cross-coupling is clearly revealed.

Fig. 4,5,6
here

The cyclobutane ring can adopt a wide range of conformations, depending on the substituents present. While $^3J_{cis}$ is generally larger

than $^3J_{\text{trans}}$, the ranges that these coupling constants can assume have considerable overlap ($^3J_{\text{cis}} = 4.6-11.4$ Hz, $^3J_{\text{trans}} = 2.24-10.7$ Hz) (Fleming & Williams, 1967). The assignment of cis or trans stereochemistry cannot therefore be made on the basis of evidence derived solely from coupling constants. Nuclear Overhauser effect (NOE) experiments, however, are ideally suited for distinguishing such stereochemistries. For a proton-proton spin system, the principal mode of relaxation is through a dipole-dipole interaction that has a r^{-6} dependence on the spatial separation of the observed and irradiated spins (Bell & Saunders, 1969). The cyclobutyl systems considered in this study are ideally suited for NOE enhancement experiments since the molecule is fairly rigid in the region of interest, minimizing complications due to rotation about carbon-carbon bonds. In addition, proton-proton and proton-methyl distances in diastereomeric cis or trans fused ring systems will be substantially different, so that the r^{-6} dependence of the NOE will lead to easily distinguishable differences. NOE enhancements are thus a sensitive indicator of relative stereochemistry in a substituted cyclobutane.

The NOE difference spectra for 8-MOP-F38 are shown in Figure 5. Irradiation of the 5'-H resonance in F38 resulted in approximately equal enhancements of the 6-H(dT) and 4'-H resonances (Table VI). This result not only corroborates the assignment of syn stereochemistry as

Table VI
here

derived above from a consideration of the coupling pattern, but also establishes that these three protons (5'-H, 4'-H and 6-H(dT)) are cis with respect to the plane of the cyclobutane ring. The 5-CH₃ of the thymidine moiety was shown to be adjacent and cis to 4'-H and 6-H(dT) when irradiation of this methyl led to a substantial enhancement of these two resonances. No significant NOE was observed between the 5-CH₃ of the thymidine moiety and 5'-H of the 8-MOP residue. These results are consistent only with a cis-syn stereochemistry, as shown in structures 3 and 3'. Virtually identical NOE results were obtained with 8-MOP-F37, indicating that these two adducts must have the stereochemical relationship shown in structures 3 and 3'. Additional support for the existence of cis-syn stereochemistry was obtained from NOE enhancement measurements on the aglycone derivative of F37 and F38, 8-MOP-F40H⁺ (Table VI).

Struc.
3, 3', 4
here

The ¹H-NMR data obtained on F36 are summarized in Table V. These data are consistent with its structural assignment as a furan-side psoralen-deoxyuridine adduct, having cis-syn stereochemistry. In this case there are four protons about the cyclobutane ring, resulting in four multiplets. There are two sets of approximately equal coupling constants, with $J_{4',5'} = J_{4',5} = 5.5$ Hz and $J_{5',6} = J_{5,6} = 9.2$ Hz. The small quantities of F36 available for analysis made it difficult to accurately measure the values of the ⁴J cross-couplings, although

an upper limit of 1.5 Hz was established. As was found for F37 and F38, there is an additional complication due to coupling between the 4'-H and 5-H of the psoralen. There was insufficient material to attempt NOE enhancement experiments on this adduct.

The $^1\text{H-NMR}$ spectrum of 8-MOP-F41 confirmed its assignment as a pyrone-side monoadduct to thymidine (Table V and Figure 6). The psoralen protons at 4' and 5' are present in the aromatic region, while the 3-H and 4-H have undergone upfield shifts into the 3.4-4.5 ppm region. The 3-H is coupled to both 4-H and 6-H of thymidine, resulting in a doublet of doublets at 3.56 ppm. The two cyclobutyl doublets collapse to singlets when the 3-H is irradiated. The regiochemistry of this system is therefore syn, as shown in structures 7 and 9. The signal to noise ratio for this sample was not high enough to obtain reliable NOE enhancements. While this does not allow an unambiguous assignment of cis or trans stereochemistry, the cis configuration (structure 7) is most consistent with the proposed model for psoralen-DNA interactions (Straub et al, 1981). The $^1\text{H-NMR}$ spectrum of 8-MOP-F42H⁺ (Table V) is consistent with its assignment as an aglycon derivative of 8-MOP-F41.

TMP-DNA Adducts. The $^1\text{H-NMR}$ spectrum of TMP-F49 is shown in Figure 7. Obviously there are more resonances present in this spectrum than are required by a single TMP-nucleoside adduct. The mass spectral results

Struc.
5,6,7,8
9,10
here

Fig. 7
here

clearly indicate the presence of a TMP-thymidine adduct. Most of the proton and methyl resonances occur in pairs with slight chemical shift differences, suggesting the presence of two diastereomers in approximately a 1/1 ratio. The 4'-H and 5'-CH₃ resonances of the TMP moiety along with the 5-CH₃ and 6-H resonances of the thymidine are shifted to higher field, indicating that cycloaddition has occurred on the furan side of the psoralen.

No vicinal (³H) coupling of the two cyclobutane protons was observed for either diastereomer, reflecting a syn orientation. There is cross-coupling ($J_{4',6} \sim 2$ Hz) of the diagonally positioned 6-H(dT) and 4'-H(TMP), analogous to a situation observed for the 8-MOP-thymidine adducts. In order to clearly reveal the cross-coupling between the two cyclobutane protons, it was necessary to decouple the 5-H of the TMP moiety. Although there was insufficient TMP-F49 to provide the signal to noise ratio necessary for NOE enhancement measurements, enough of the aglycone derivative of F49 for NOE analysis was obtained by acid hydrolysis of a sample of TMP-modified DNA (TMP-F50H⁺). Irradiation of either of the two cyclobutyl methyls (5'-CH₃ and 5-CH₃(dT)) showed relatively large and approximately equal enhancements for both of the cyclobutyl protons (Table VII). This result is consistent only with a structure where both methyls are cis and adjacent to both cyclobutyl protons. The aglycone of TMP-F49 (and therefore F49) thus has the same cis-syn stereochemistry

Table VII
here

as found for the 8-MOP-nucleoside adducts.

TMP-F47 had a $^1\text{H-NMR}$ spectrum consistent with its assignment as a TMP-deoxyuridine adduct. There are three methyl resonances, corresponding to the CH_3 's at C-4, C-8, and C-5' of TMP. The cyclobutyl system consists of three protons and the 5'- CH_3 from the TMP moiety. The 5-H resonance of the deoxyuridine is an apparent triplet, indicating that $J_{4',5} = J_{5,6}$. Irradiation at 5-H of the deoxyuridine results in a pair of apparent singlets for 4'-H of the TMP and 6-H of the pyrimidine residue; this implies that cross coupling between these diagonally positioned protons is less than the 1.5 Hz line width of the signals. These results are consistent with *cis-syn* stereochemistry. In addition, F47 as isolated appears to be a single diastereomer.

Discussion

The photocycloaddition of a pyrimidine to a psoralen generates a cyclobutane ring with four asymmetric centers. In nucleoside adducts, an additional invariant chiral center is present in the deoxyribose moiety. Reaction can take place at either the 4',5'-furan or 3,4-pyrone side double bonds of the psoralen. This addition can occur in either a *syn* or *anti* configuration, where *syn* refers to a configuration where the 5-position of the pyrimidine and the 5' or 3-position of the psoralen are at diagonally opposed corners of the cyclobutane ring.

Anti, then, refers to a configuration where the 5-position of the pyrimidine is adjacent to either the 5'- or 3-position of the psoralen. There are thus four sets of diastereomers (furan-side syn, furan-side, anti, pyrone-side syn, pyrone-side anti), each having in theory $2^4 = 16$ possible isomers. Twelve of the sixteen isomers will incorporate a trans-fused cyclobutane. While such a trans-fused four-membered ring is possible, the steric strain present in a tricyclo[4.0.0.3] system incorporating one or two trigonal centers would be prohibitive. As discussed below, NOE enhancement measurements are an effective tool for unequivocally assigning stereochemistry in such systems.

The four remaining isomers which incorporate a cis-fused cyclobutane are in fact two pairs of enantiomers if one disregards the presence of the deoxyribose. This can occur at each of the possible addition sites on the psoralen, so that there are a total of 8 possible pairs of enantiomers. The notation cis or trans is used to refer to whether the pyrimidine and psoralen moieties are on the same or opposite sides of the cyclobutane ring. These 8 possible isomers are denoted as structures 3-10. Each structure has an accompanying diastereomer, where the configuration at each asymmetric center on the cyclobutane ring is opposite that shown. The presence of an additional invariant asymmetric center in the deoxyribose moiety of nucleoside adducts yields a pair of diastereomers rather than enantiomers. Removal of the deoxy-

ribose will result in pairs of enantiomers. The relationship between these diastereomeric pairs is illustrated for the cis-syn isomers, structures 3 and 3'.

The mass spectral and $^1\text{H-NMR}$ results outlined above indicate that both 8-MOP and TMP undergo photoreaction with DNA to yield as major products a pair of diastereomeric furan-side thymidine adducts. The two psoralens also undergo reaction with deoxycytidine (and perhaps deoxy-5-methylcytosine) residues. The NMR results unequivocally define the stereochemistry of these adducts as cis-syn furan-side adducts, where cycloaddition has occurred between the 4',5' double bond of the psoralen and the 5,6 double bond of the pyrimidine. The pairs of 8-MOP-thymidine (F37, F38) and TMP-thymidine (F49) adducts therefore have the stereochemical relationship represented in structures 3 and 3'. That is, the stereochemistry at each asymmetric center of the cyclobutane ring is opposite and equal, but the presence of an invariant chiral center in the deoxyribose results in a pair of diastereomers.

Removal of the deoxyribose can be accomplished by acid hydrolysis, since reduction of the 5,6-double bond in pyrimidine nucleosides greatly increases the acid lability of the C-N glycosidic bond. When acid hydrolysis is carried out on 8-MOP-F37 and -F-38, they are converted to thymine-8-MOP adducts which co-elute under our standard HPLC conditions. The two hydrolysis products have opposite and equal

circular dichroism spectra, as expected for a pair of enantiomers. The CD spectra of F37 and F38 are also approximately opposite and equal (Figure 8), since the deoxyribose moiety influences the overall molar ellipticity of each diastereomer to a small degree. Similar results were obtained for TMP-F49. The $^1\text{H-NMR}$ spectrum of TMP-F49 indicates that it is actually a 1/1 mixture of diastereomeric adducts, each having cis-syn stereochemistry. Acid hydrolysis of this mixture results in apparently a single product by NMR, as expected for an enantiomeric mixture.

The formation of diastereomeric pairs of adducts can be explained by considering the steps involved in the binding of psoralens to DNA. The formation of a non-covalent complex occurs by intercalation of the psoralen between adjacent pairs of the DNA double helix (Wiesehahn & Hearst, 1978). This complex can then undergo photoreaction to yield a covalent nucleoside-psoralen adduct. The question of which of the two possible cis-syn diastereomers is formed is determined by whether the psoralen is on top, or underneath, of a given base pair.

The deoxyuridine adducts found for the two psoralens (8-MOP-F36 and TMP-F47) appear to be single cis-syn diastereomers based on their $^1\text{H-NMR}$. It is possible that their accompanying diastereomers are present as minor uncharacterized products in our chromatographic system. Also, 8-MOP-F41 may consist of more than a single diastereomer, but complete

analysis of this fraction is complicated by the facile hydrolysis of the lactone in 3,4-dihydropсорalen derivatives, leading to multiple products. An accurate measurement of the relative amounts of the two diastereomeric adducts (8-MOP-F37, -F38) formed in the reaction with double-stranded DNA is important, since a preference for the formation of one diastereomer indicates a preference by the psoralen for intercalation on top, or underneath, of an AT base pair. Such short range sequence specificity has been observed with other intercalating agents such as ethidium bromide (Kastrup et al, 1978).

The results in Table I indicate a preference in the diastereomeric ratio of F38/F37 of 7/4. Confirmation of this preference has been obtained by hydrolyzing 8-MOP-modified DNA and measuring the CD of the isolated 8-MOP-thymine adduct, 8-MOP-F40H⁺. The CD was compared to the CD spectrum of the pure enantiomer obtained by acid hydrolysis of 8-MOP-F37 and -F38. In contrast, the two TMP-thymidine adducts present in TMP-F49 appear to be formed in a 1/1 ratio. The absolute configurations of these adducts are currently under investigation, and will allow us to assign the sequence specificity of 8-MOP as (5')TpX or (5')Xpt.

The results obtained in this study indicate that complete hydrolysis of psoralen-modified DNA can be achieved, either enzymatically or by treatment with mild acid. Characterization of the psoralen-modified

DNA by HPLC will then allow direct quantification of the number and type of psoralen-nucleoside monoadducts and crosslinks present in the DNA. This methodology has the potential of becoming a standard assay for the detection of both mono- and di-adducts in psoralen-modified nucleic acids.

The results of the present study are similar to those obtained previously by us with a synthetic psoralen derivative, 4'-hydroxymethyl-4,5',8-trimethylpsoralen (HMT) (Straub et al, 1981). The photoreaction of HMT with double-stranded DNA yields a set of psoralen-nucleoside monoadducts similar to those observed for TMP. The major products are diastereomeric cis-syn, furan-side adducts with thymidine and deoxyuridine (deoxycytidine) residues. With both HMT and TMP, addition to the 3,4-(pyrone) double bond of the psoralen is a minor reaction compared to addition to the 4',5'-(furan) double bond. In contrast, the reaction of 8-MOP with DNA results in a substantial amount (18%) of the pyrone-side adduct, F41. These results do not agree with predictions of the sites of reactivity in psoralens made by others (Song et al 1971; Chatterjee & Cantor, 1978).

A possible explanation for the observed specificity is that the steric constraints imposed by the noncovalent psoralen-DNA intercalation complex prior to irradiation in part determines the stereochemistry of the products. A model for such an intercalation complex has been proposed, where photoreaction with adjacent pyrimidine residues

will lead to the observed furan- or pyrone-side, cis-syn stereochemistry at either the 4',5'- or 3,4-double bonds of the psoralen. It is possible that a methyl substituent at either the 3 or 4-positions of the psoralen leads to unfavorable steric interactions with the 5-methyl of an adjacent thymine, resulting in a decreased yield of pyrone-side adducts. This interaction is absent for 8-MOP, and could account for the increased yield of pyrone-side adduct observed with this psoralen. Other factors such as electronic effects and solubility considerations will of course influence the overall quantum yield of the psoralen-DNA reaction. This model also predicts that cis-syn stereochemistry is in part the result of carrying out the photoreaction with native, double-stranded DNA. It is possible that other diastereomeric adducts could result from reaction with alternative nucleic acid secondary or tertiary structures. Should this be the case, then psoralens could be uniquely useful as probes of secondary and tertiary nucleic acid structure. Experiments are currently underway to test this hypothesis.

Conclusions

We have isolated and characterized four mononucleoside adducts from the photoreaction of 8-MOP or TMP with native, double-stranded DNA. Two diastereomeric adducts of thymidine are formed by cycloaddition to the 4',5'-(furan) double bond of the psoralen, with cis-

syn stereochemistry. Small amounts of a cis-syn deoxycytidine adduct are also formed, which undergoes hydrolytic deamination to a deoxyuridine product. In the case of 8-MOP, substantial quantities of a pyrone-side adduct are formed, where cycloaddition has taken place at the 3,4-double bond of the psoralen. The photoreaction between TMP and DNA results in only minor quantities of a deoxythymidine-pyrone side adduct. All of the adducts have cis-syn stereochemistry, determined in part by the constraints imposed by the DNA helix on the geometry of the non-covalent intercalation complex formed by the psoralen and DNA prior to irradiation. Acid hydrolysis studies on 8-MOP or TMP-modified DNA resulted in the isolation and characterization of a mixture of enantiomeric cis-syn thymine-psoralen adducts, identical to the aglycone derivatives of the deoxythymidine adducts. With double-stranded DNA there is a 7/4 preference for the formation of one diastereomeric nucleoside 8-MOP adduct, indicating a specificity by the psoralen for a (5')TpX or (5')XpT sequence. The methodology described in this study provides an assay for the complete characterization of both mono- and di-adducts in psoralen-modified nucleic acids.

Acknowledgment

This research was supported in part by the Office of Environmental Research and Development, Biomedical and Environmental Research Division of the DOE under Contract No. W-7405-ENG-48, and the National Institute of General Medical Sciences (Grant GM-25151).

NMR studies were carried out at the University of California, Davis, NMR Facility under the auspices of NSF Grant No. CHE 79-04832, or at the Stanford Magnetic Resonance Laboratory, under the auspices of NSF and NIH Grants Nos. GP-263633 and RR-00711. The authors thank Dr. J. Dallas and A. Ribiero for their assistance in obtaining NMR data. Mass spectrometry studies were carried out at the University of California, Berkeley, Bioorganic, Biomedical Mass Spectrometry Resource supported by NIH Grant RR-00719, Dr. A. L. Burlingame, Director.

References

- Ashwood-Smith, M. J., Poulton, G. A., Barker, M., & Mildenburger, M. (1980) Nature 285, 407-409.
- Beckey, O. H. (1977) Principles of Field Ionization and Field Desorption Mass Spectrometry pp 64-72, Pergamon Press, Oxford.
- Bell, R. A., & Saunders, J. K. (1969) Can. J. Chem. 48, 1114-1122.
- Burlingame, A. L., Olsen, R. W., & McPherron, R. (1974) in Advances in Mass Spectrometry (West, A. R., ed.) Vol. 6, pp 1053-1059, Applied Science Publishers, Barking, Essex.
- Chatterjee, P. K., & Cantor, C. R. (1978) Nucl. Acids Res. 5, 3619-
- Cole, R. S. (1970) Biochim. Biophys. Acta 217, 30-39.
- Cole, R. S. (1971) Biochim. Biophys. Acta 254, 30-39.
- Ehrlich, M., & Wang, R.Y.-H. (1981) Science 212, 1350-1357.
- Epstein, J. E. (1979) New Engl. J. Med. 300, 852-853.
- Fleming, I., & Williams, D. H. (1967) Tetrahedron 23 2747-2765.
- Freeman, K. B., Hariharan, P. V., & Johns, H. E. (1965) J. Mol. Biol. 13, 833.
- Hanson, C. V., Shen, C.-K. J., & Hearst, J. E. (1976) Science 193, 62-64.
- Harten, M. L., Felkner, I. C., & Song, P.-S. (1976) Photochem. Photobiol. 24, 491.
- Isaacs, S. T., Rapoport, H., & Hearst, J. E. (1981) J. Lab. Comp. Radiopharm., in press.

- Kastrup, R. V., Young, M. A., & Krugh, T. R. (1978) Biochemistry 17, 4855-4864.
- McCloskey, J. A. (1974) in Basic Principles in Nucleic Acid Chemistry (Ts'o, P. O. P., ed) Vol. 1, pp 220-227, Academic Press, New York.
- Rodighiero, G., Musajo, L., Dall'Acqua, F., Marciani, S., Caporale, G., & Ciavatta, L. (1970) Biochim. Biophys. Acta 217, 40.
- Scott, B. R., Pathak, M. A., & Mohn, G. R. (1976) Mutation Res. 39, 29-74.
- Shen, C.-K. J., & Hearst, J. E. (1977) J. Mol. Biol. 112, 495-507.
- Song, P.-S., Harter, M. I., Moore, T. A., & Herndon, W. C. (1971) Photochem. Photobiol. 14, 521-
- Song, P.-S., & Tapley, K. J. (1979) Photochem. Photobiol. 29, 1177-1197.
- Stern, R. S., Thibodeau, L. A., Kleinerman, R. A., Parrish, J. A., & Fitzpatrick, T. B. (1979) New Engl. J. Med. 300, 809-813.
- Straub, K., Kanne, D., Hearst, J. E., & Rapoport, H. (1981) J. Am. Chem. Soc., 103, 2347-2355.
- Wiesehahn, G., & Hearst, J. E. (1978) Proc. Natl. Acad. Sci. USA 75, 2703-2708.
- Wiesehahn, G. P., Hyde, J. E. & Hearst, J. E. (1977) Biochemistry 16, 925-932.
- Wolff, K., Gschnait, F., Honigsman, H., Konrad, K., Parrish, J. A., & Fitzpatrick, T. B. (1977) Br. J. Dermatol. 96, 1-10.

Table I. ^3H Recoveries and Modification Levels for Psoralen-DNA Reactions

	<u>8-MOP^a</u>	<u>TMP^b</u>
1. Psoralen added per base pair	0.28	0.50
2. Psoralen bound per base pair	0.065	0.110
3. HPLC fractions, enzyme-hydrolyzed psoralen-DNA, % ^c bound ^3H in	<u>fraction %</u>	<u>fraction %</u>
	F33 26	F34-36 ^d 41
	F36 2	F47 ^e <2
	F37 16	F49 ^f 55
	F38 28	
	F42 18	F52 ^g 2.5

^a Single addition of psoralen, 2.5 hr irradiation. ^b Multiple additions of psoralen, 10 min irradiation after each addition. ^c Average of three determinations. ^d Nucleoside-psoralen-nucleoside cross-linked fraction. ^e Furan-side deoxyuridine monoadducts. ^f Diastereomeric furan-side deoxythymidine monoadducts. ^g Pyrone-side deoxythymidine monoadducts.

Table II. High Resolution Mass Spectral Data for 8-MOP-Nucleoside Adducts.

F36 ^a		F37 ^b	F38 ^b	F41		
m/z	composition	m/z	m/z	composition ^c	m/z	composition
573.1736	C ₂₀ H ₁₅ O ₉ N ₂ TMS ₂	587.1843	587.1843	C ₂₁ H ₁₇ O ₉ N ₂ TMS ₂	386.1730	C ₁₀ H ₁₂ N ₂ O ₅ TMS ₂
543.1604	C ₁₉ H ₁₃ O ₈ N ₂ TMS ₂	557.1749	557.1749	C ₂₀ H ₁₅ O ₈ N ₂ TMS ₂	378.1375	C ₁₂ H ₈ O ₅ TMS ₂
529.1467	C ₁₈ H ₁₁ O ₈ N ₂ TMS ₂	543.1608	543.1617	C ₁₉ H ₁₃ O ₈ N ₂ TMS ₂	363.1094	C ₁₁ H ₅ O ₅ TMS ₂
501.1490	C ₁₇ H ₁₁ O ₇ N ₂ TMS ₂	515.1608	515.1657	C ₁₈ H ₁₃ O ₇ N ₂ TMS ₂	289.0884	C ₁₂ H ₈ O ₄ TMS
483.1211	C ₂₀ H ₁₄ O ₈ N ₂ TMS	497.1376	497.1381	C ₂₁ H ₁₆ O ₈ N ₂ TMS	216.0417	C ₁₂ H ₈ O ₄
455.1268	C ₁₉ H ₁₄ O ₇ N ₂ TMS	469.1436	469.1436	C ₂₀ H ₁₆ O ₇ N ₂ TMS	201.0184	C ₁₁ H ₅ O ₄
427.1328	C ₁₈ H ₁₄ O ₆ N ₂ TMS	441.1486	441.1484	C ₁₉ H ₁₆ O ₆ N ₂ TMS	188.0467	C ₁₁ H ₈ O ₃
401.1151	C ₁₆ H ₁₂ O ₆ N ₂ TMS	415.1328	415.1327	C ₁₇ H ₁₄ O ₆ N ₂ TMS	173.0245	C ₁₀ H ₅ O ₃
355.0947	C ₁₈ H ₁₅ O ₆ N ₂	369.1109	369.1094	C ₁₉ H ₁₇ O ₆ N ₂	126.0433	C ₅ H ₆ N ₂ O ₂
329.0781	C ₁₆ H ₁₃ O ₆ N ₂	343.0923	343.0904	C ₁₇ H ₁₅ O ₆ N ₂		
289.0887	C ₁₂ H ₈ O ₄ TMS	289.0900	289.0894	C ₁₂ H ₈ O ₄ TMS		
216.0429	C ₁₂ H ₈ O ₄	216.0416	216.0431	C ₁₂ H ₈ O ₄		
201.0182	C ₁₁ H ₅ O ₄	201.0185	201.0185	C ₁₁ H ₅ O ₄		
188.0475	C ₁₁ H ₈ O ₃	188.0472	188.0473	C ₁₁ H ₈ O ₃		
173.0240	C ₁₀ H ₅ O ₃	173.0236	173.0242	C ₁₀ H ₅ O ₃		
99.0261	C ₄ H ₇ OSi	113.0422	113.0425	C ₅ H ₉ OSi		

^a FDMS of 8-MOP-F36 gave m/z 216, 444 (M⁺), 467 (M + Na⁺). ^b FDMS of 8-MOP-F37, F38 gave m/z 216, 458 (M⁺), 481 (M + Na⁺), 939 (M₂ + Na⁺). ^c all errors for elemental composition matches are <10 ppm.

Table III. High Resolution Mass Spectral Data for TMP-Nucleoside Adducts.

TMP-F49 ^a		TMP-F47 ^b		TMP-F52		assignment
m/z	elemental composition	m/z	elemental composition	m/z	elemental composition	
599.2240	C ₂₃ H ₂₁ O ₈ N ₂ TMS ₂	585.2107	C ₂₂ H ₁₉ O ₈ N ₂ TMS ₂	-	-	M - CH ₃ (ion "m")
569.2160	C ₂₂ H ₁₉ O ₇ N ₂ TMS ₂	555.2009	C ₂₁ H ₁₇ O ₇ N ₂ TMS ₂	-	-	m - CH ₂ O
555.1997	C ₂₁ H ₂₀ O ₇ N ₂ TMS ₂	541.1842	C ₂₀ H ₁₅ O ₇ N ₂ TMS ₂	-	-	
509.1733	C ₂₃ H ₂₀ O ₇ N ₂ TMS	495.1590	C ₂₂ H ₁₈ O ₇ N ₂ TMS	-	-	m - HOTMS
481.1781	C ₂₂ H ₂₀ O ₆ N ₂ TMS	467.1652	C ₂₁ H ₁₈ O ₆ N ₂ TMS	-	-	
453.1822	C ₂₁ H ₂₀ O ₆ N ₂ TMS	439.1688	C ₂₀ H ₁₈ O ₅ N ₂ TMS	-	-	M - C ₃ H ₄ O ₃ TMS
427.1687	C ₁₉ H ₁₈ O ₅ N ₂ TMS	413.1540	C ₁₈ H ₁₆ O ₅ N ₂ TMS	-	-	
-	-	-	-	390.1685	C ₁₄ H ₁₂ O ₄ TMS ₂	TMP, di TMS ether/ester
-	-	-	-	386.1726	C ₁₀ H ₁₂ O ₅ N ₂ TMS ₂	thymidine, diTMS ether
381.1469	C ₂₁ H ₂₁ O ₅ N ₂	367.1281	C ₂₀ H ₁₉ O ₅ N ₂	-	-	M - C ₃ H ₃ O ₃ TMS ₂
-	-	-	-	375.1473	C ₁₃ H ₉ O ₄ TMS ₂	390 - CH ₃
367.1310	C ₂₀ H ₁₉ O ₅ N ₂	353.1140	C ₁₉ H ₁₇ O ₅ N ₂	-	-	
355.1279	C ₁₉ H ₁₉ O ₅ N ₂	341.1136	C ₁₈ H ₁₇ O ₅ N ₂	516.2140	C ₁₉ H ₁₈ O ₆ N ₂ TMS ₂	M - dr
301.1236	C ₁₄ H ₁₂ O ₃ TMS	301.1257	C ₁₄ H ₁₂ O ₃ TMS	301.1262	C ₁₄ H ₁₂ O ₃ TMS	TMP + TMS
228.0787	C ₁₄ H ₁₂ O ₃	228.0782	C ₁₄ H ₁₂ O ₃	228.0786	C ₁₄ H ₁₂ O ₃	TMP
200.0832	C ₁₃ H ₁₂ O ₂	200.0833	C ₁₃ H ₁₂ O ₂	200.0848	C ₁₃ H ₁₂ O ₂	288 - CO
-	-	-	-	199.0896	C ₅ H ₆ O ₂ N ₂ TMS	thymine + TMS
199.0759	C ₁₃ H ₁₁ O ₂	199.0753	C ₁₃ H ₁₁ O ₂	-	-	228 - HCO

^a FDMS of TMP F49 gave m/z 126, 228, 470 (M⁺), 493 (M + Na⁺), 963 (M₂Na⁺). ^b FDMS of TMP F47 gave m/z 228, 456 (M⁺), 479 (M + Na⁺), 935 (M₂Na⁺). ^c Errors for composition matches are <10 ppm. ^d dR = deoxy-ribose, TMS = trimethylsilyl.

Table IV. High Resolution Mass Spectral Data for Acid Hydrolyzed Psoralen-DNA Monoadducts

<u>8-MOP-F40H⁺ ^a</u>		<u>8-MOP-F42H⁺ ^b</u>		assignment
m/z	elemental composition ^d	m/z	elemental composition ^d	
342.0850	C ₁₇ H ₁₄ O ₆ N ₂			M ⁺
216.0425	C ₁₂ H ₈ O ₄	216.0412	C ₁₂ H ₈ O ₄	8-MOP
201.0174	C ₁₁ H ₅ O ₄	201.0182	C ₁₂ H ₈ O ₄	216 - CH ₃
188.0472	C ₁₁ H ₈ O ₃	188.0468	C ₁₁ H ₈ O ₃	216 - CO
173.0241	C ₁₀ H ₅ O ₃	173.0240	C ₁₀ H ₅ O ₃	
145.0292	C ₉ H ₅ O ₂	145.0293	C ₉ H ₅ O ₂	
140.0579	C ₆ H ₈ O ₂ N ₂	140.0581	C ₆ H ₈ O ₂ N ₂	
126.0431	C ₅ H ₆ O ₂ N ₂	126.0433	C ₅ H ₆ O ₂ N ₂	thymine
83.0372	C ₄ H ₅ ON	83.0371	C ₄ H ₅ NO	126 - HNCO
<u>TMP-F50H⁺ ^c</u>				
354.1217	C ₁₉ H ₁₈ O ₅ N ₂			M ⁺
228.0793	C ₁₄ H ₁₂ O ₃			TMP
200.0836	C ₁₃ H ₁₂ O ₂			228 - CO
199.0761	C ₁₃ H ₁₁ O ₂			228 - HCO
185.0592	C ₁₂ H ₉ O ₂			
126.0429	C ₅ H ₆ O ₂ N ₂			thymine
83.0374	C ₄ H ₅ ON			126 - HNCO

^a FDMS gives m/z 126, 216, 342 (M⁺), 365 (M + Na⁺). ^b FDMS gives m/z 126, 216.

^c FDMS gives m/z 126, 228, 354 (M⁺), 377 (M + Na⁺). ^d Errors for composition matches are <10 ppm.

Table V. Chemical Shifts (δ , ppm)^a and Coupling Constants of 8-MOP-DNA Monoadducts.

moiety ^b	F36	F37	F38	F41	F40H ⁺	F42H ⁺
C-4-H (8-MOP)	8.044 d	7.930 d	7.865 d	4.161 d	7.899 d	3.59 d
$J_{4,3}$	9.5 Hz	9.5 Hz	9.5 Hz	9.7 Hz	9.5 Hz	10 Hz
C-5-H (8-MOP)	7.181 s	7.182 s	7.103 s	7.284 s	7.131 s	7.220 s
C-3-H (8-MOP)	6.339 d	6.343 d	6.294 d	3.54 dd	6.317 d	3.499 dd
C-1'-H (dR)	6.208 t	6.200 t	6.210 t	6.280 dd	-	-
C-5'-H (8-MOP)	5.569 t	5.546 dd	5.600 t	7.603 d	5.581 t	7.720 d
$J_{5',4'}$	5.4 Hz	5.5 Hz	5.5 Hz	2.2 Hz	5.5 Hz	$J_{4',5'} = 2.2$ Hz
$J_{5',6}$	9.2 Hz	5.8 Hz	5.5 Hz	-	5.5 Hz	-
C-6-H (U or T)	4.519 m	4.435 dd	4.480 dd	4.255 d	4.155 dd	4.015 d
$J_{4',6}$	1.5 Hz	1.2 Hz	1.5 Hz	-	2.5 Hz	$J_{3,6} = 10$ Hz
$J_{5,6}$	9.2 Hz	-	-	-	-	$J_{4,6} = 2$ Hz
C-3'-H (dR)	4.215 m	4.385 m	4.390 m	4.358 m	-	-
C-4'-H (8-MOP)	4.773 m	4.189 d	4.161 d	6.753 d	4.190 d	6.640
$J_{4',5}$	5.4 Hz	-	-	-	-	-
C-5-H (U)	4.011 t	-	-	-	-	-
C-4'-H (dR)	3.853 m	3.931 m	3.900 m	3.901 m	-	-
C-8-OCH ₃	3.894 s	3.946 s	3.880 s	4.032 s	3.963 s	3.899 s
C-5'-H (dR)	3.710 m	3.757 m	3.700 m	3.714 m	-	-
C-2',2''-H (dR)	2.103, 1.958 m	2.450, 2.300 m	2.232 m	2.08 m	-	-
C-5-CH ₃ (T)	-	1.669 s	1.655 s	1.605 s	1.655 s	1.540 s

^a δ H₂O = 4.75 ppm. ^b dR = deoxyribose; U = uracil; T = thymine.

Table VI. Nuclear Overhauser Effect (NOE) Experiments on Psoralen-DNA Monoadducts.

psoralen	irradiated resonance	observed resonance	% enhancements ^a		F40H ⁺ c
			F37 ^b	F38 ^b	
8-MOP	C-5'-H (8-MOP)	C-4'-H (8-MOP)	8.7	4.5	8.2
		C-6-H (T)	10	4.7	4.6
		C-5-CH ₃ (8-MOP)	0	0	0
	C-5-CH ₃ (T)	C-4'-H (8-MOP)	13	15	11.2
		C-6-H (T)	9.4	10.6	10.4
		C-5'-H (8-MOP)	0	0	0
TMP	C-5'-CH ₃ (TMP)	C-4'-H (TMP)			F50H ⁺ c
		C-6-H (T)			16.7
	C-5-CH ₃ (T)	C-4'-H (TMP)			15.6
		C-6-H (T)			17.8
					14.8

^a Error is $\pm 1\%$ for F38, F40H⁺, F50H⁺; $\pm 2\%$ for F37. ^b Determined in D₂O.

^c Determined in d₆-DMSO.

Table VII. Chemical Shifts (δ , ppm)^a and Coupling Constants of TMP-DNA Monoadducts

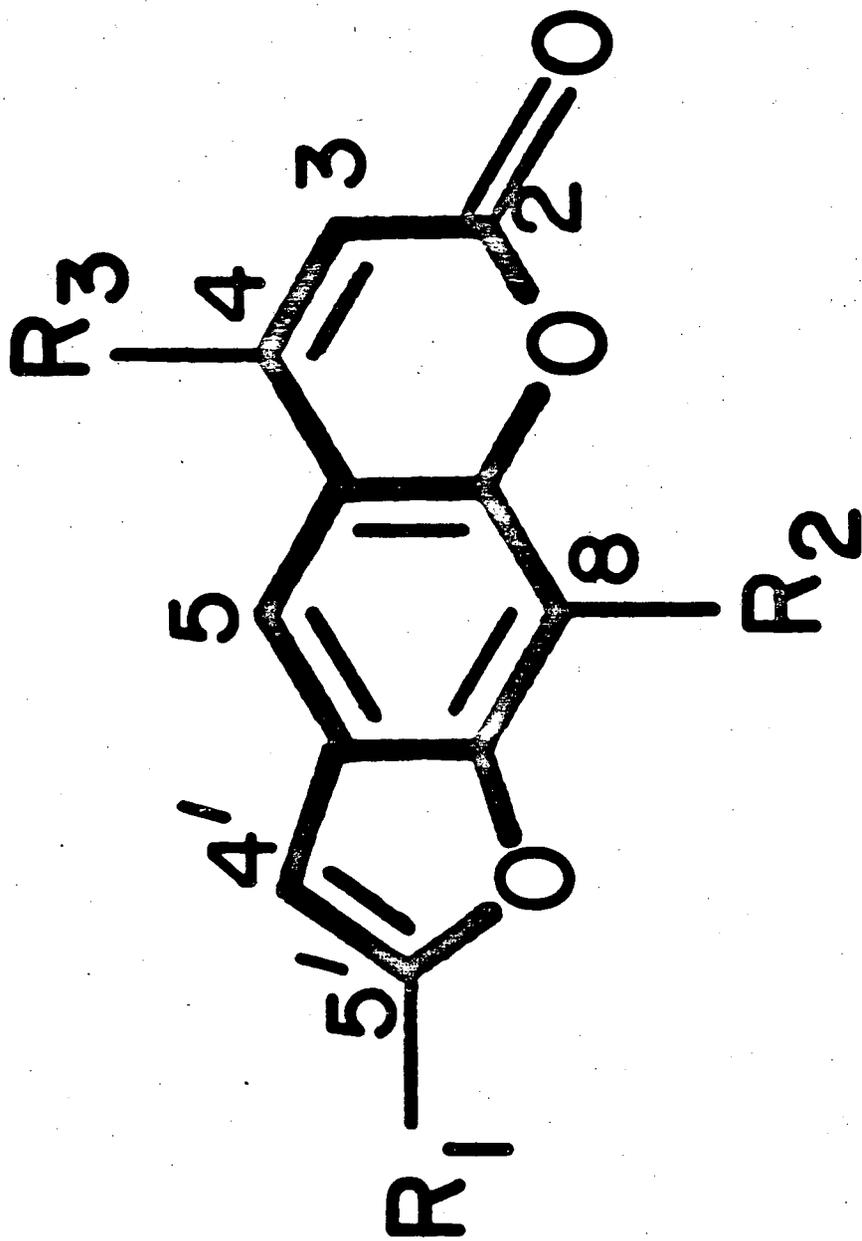
moiety ^b	F47	F49 (49 + 49')	F50H ⁺
C-5-J (TMP)	7.378 s	7.379 s	7.350 s
C-3-H (TMP)	6.249 s	6.233 s	6.279 s
C-1'-H (dR)	6.134 t	6.144, 5.954 t	-
C-6-H (U)	4.560 m	-	-
	$J_{5,6}=9.2$ Hz		
	$J_{4',6}$ 1.5 Hz		
C-3'-H (dR)	4.443 m	4.450, 4.340 m	-
C-6-H (T)	-	4.281, 4.200 d	3.824 d
		$J_{4',6}=2$ Hz	$J_{4',6}=2$ Hz
C-4'-H (TMP)	4.285 m	3.880, 3.861 dd	3.796 dd
	$J_{4',5}=9.2$ Hz	$J_{4',6}=2$ Hz	$J_{4',6}=2$ Hz
	$J_{4',6}$ 1.5 Hz	$J_{4',5}(\text{TMP})$ 1 Hz	$J_{4',5}(\text{TMP})$ 1 Hz
C-5-H (U)	4.041 t	-	-
	$J_{5,6}=J_{4',5}=9.2$ Hz		
C-4'-H (dR)	3.922 m	3.920 m	-
C-5'-H (dR)	3.722 m	3.750 m	-
C-4-CH ₃ (TMP)	2.432 s	2.410 s	2.427 s
C-2',2''-CH ₃ (dR)	2.287 m	2.280 m	-
C-8-CH ₃ (TMP)	2.246 s	2.226, 2.208 s	2.244 s
C-5'-CH ₃ (TMP)	1.771 s	1.775, 1.702 s	1.703 s
C-5-CH ₃ (T)	-	1.685, 1.602 s	1.684 s

^a $\delta_{\text{HDO}} = 4.75$ ppm. ^b TMP = 4,5',8-trimethylpsoralen; dR = deoxyribose;

U = uracil; T = thymine.

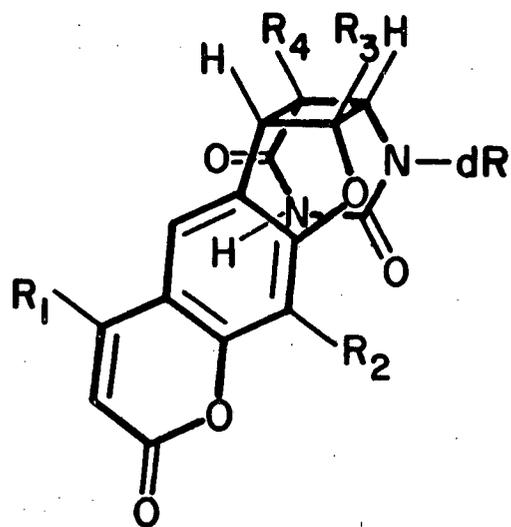
Figure Captions

- Figure 1: HPLC elution profile of (a) enzyme hydrolyzed 8-MOP modified DNA, (b) enzyme hydrolyzed TMP modified DNA.
- Figure 2: Absorption spectrum of 8-MOP F38 (—) and 8-MOP F42 (---) in 50% methanol-water. The concentration of 8-MOP-F42 is approximately 1.7X that of F38.
- Figure 3: High resolution mass spectrum (m/z 400-650) of the TMS-ether derivative of 8-MOP-F38.
- Figure 4: 360 MHz ^1H -NMR spectrum of 8-MOP-F37.
- Figure 5: 360 MHz ^1H -NMR spectrum of 8-MOP-F38, and NOE difference spectra for this adduct. Arrows refer to location of decoupling field.
- Figure 6: 360 MHz ^1H -NMR spectrum of 8-MOP-F41.
- Figure 7: 360 MHz ^1H -NMR spectrum of TMP-F49.
- Figure 8: Circular dichroism spectra of 8-MOP-F37 and 8-MOP-F38.

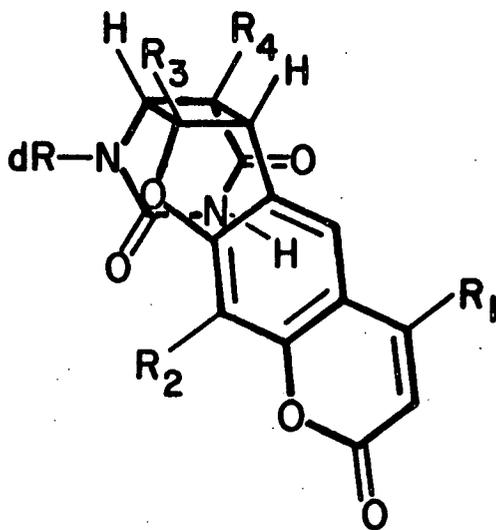


1: $R_1 = R_3 = H$, $R_2 = OCH_3$

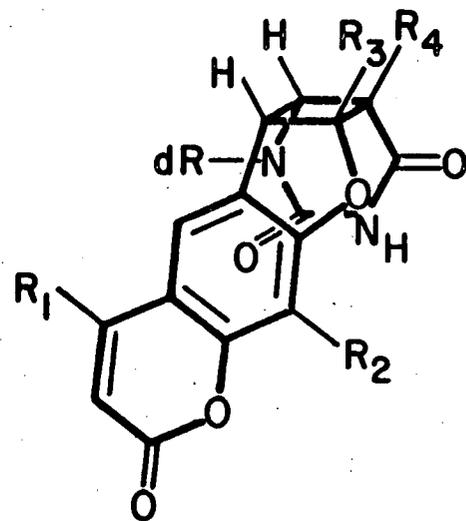
2: $R_1 = R_2 = R_3 = CH_3$



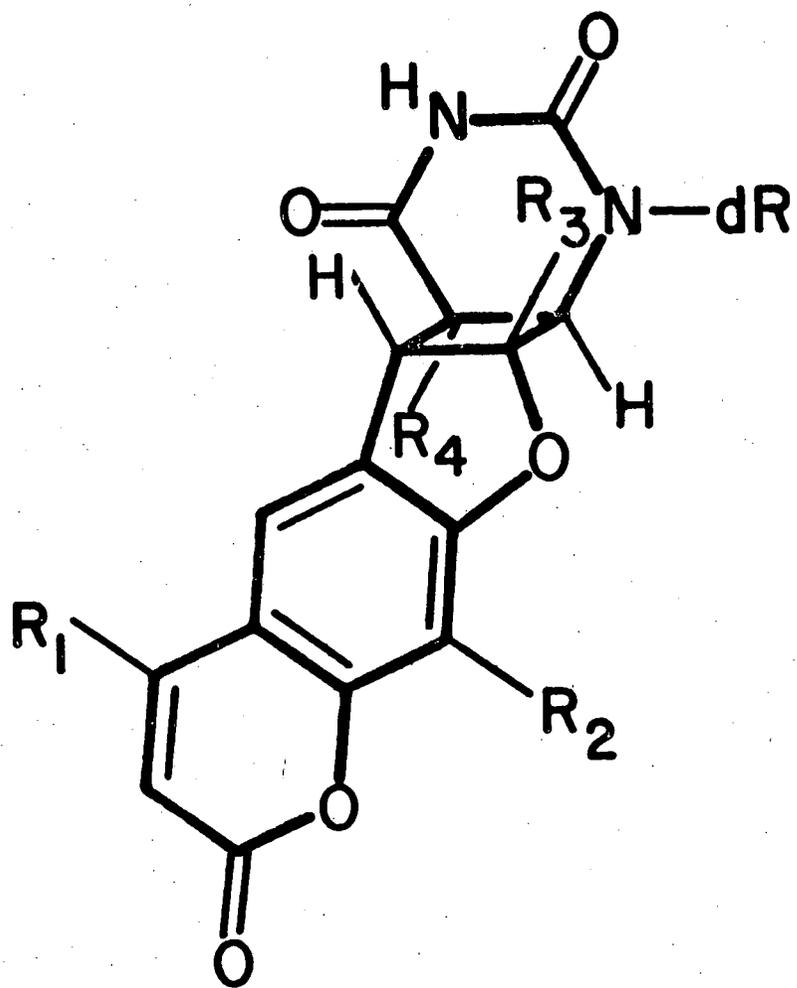
cis-syn-3



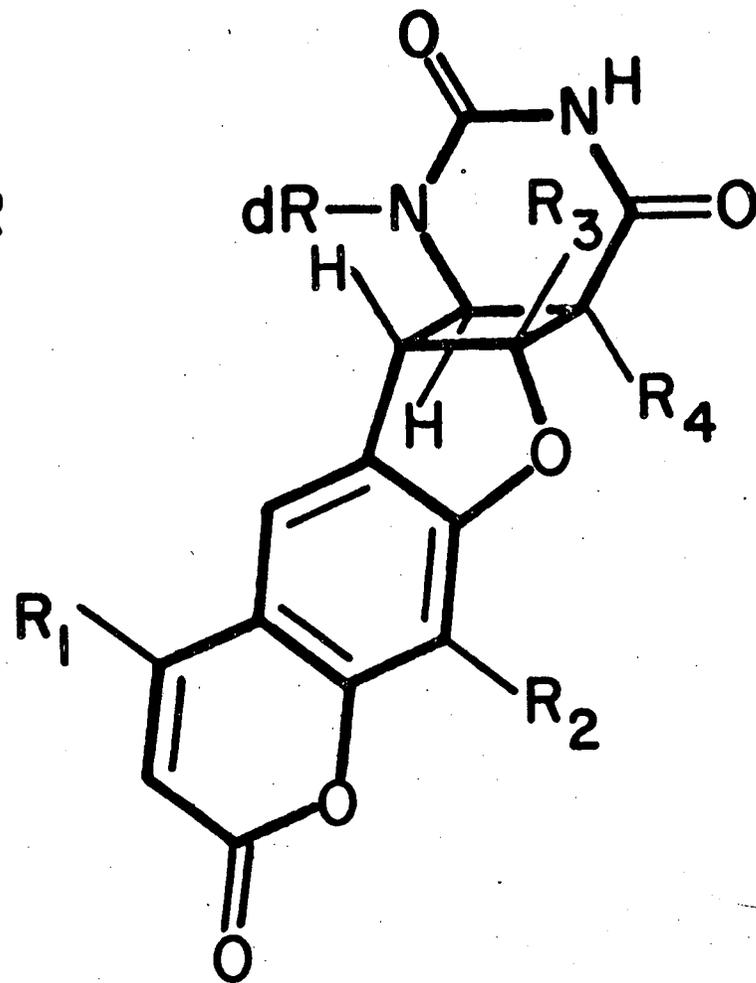
cis-syn-3'



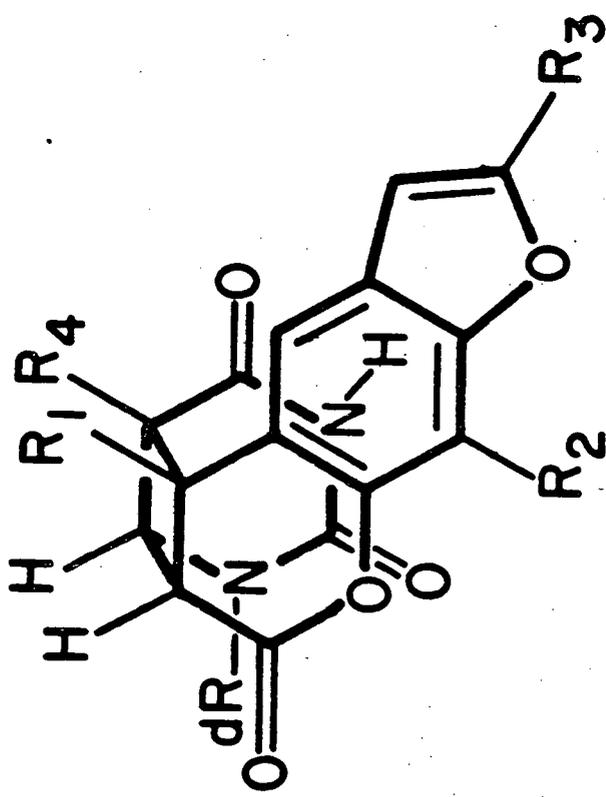
cis-syn-4



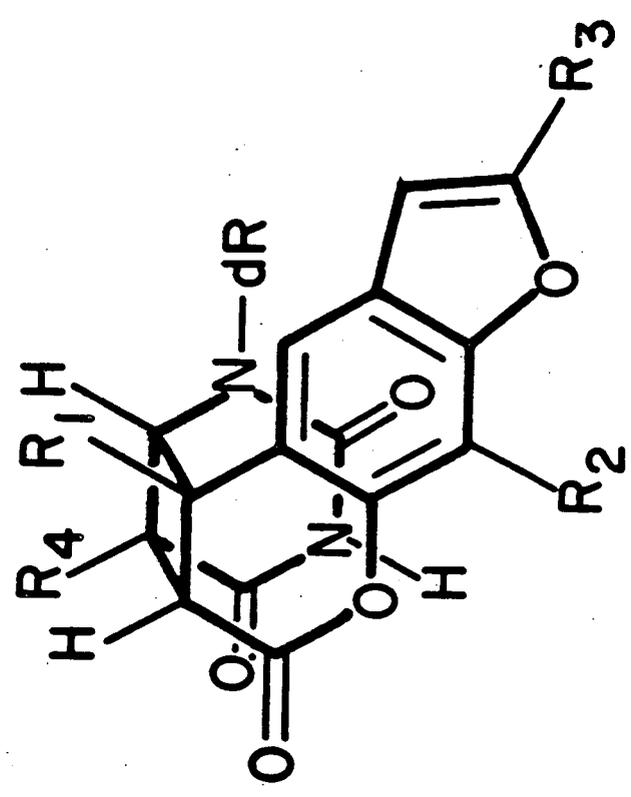
trans-syn-5



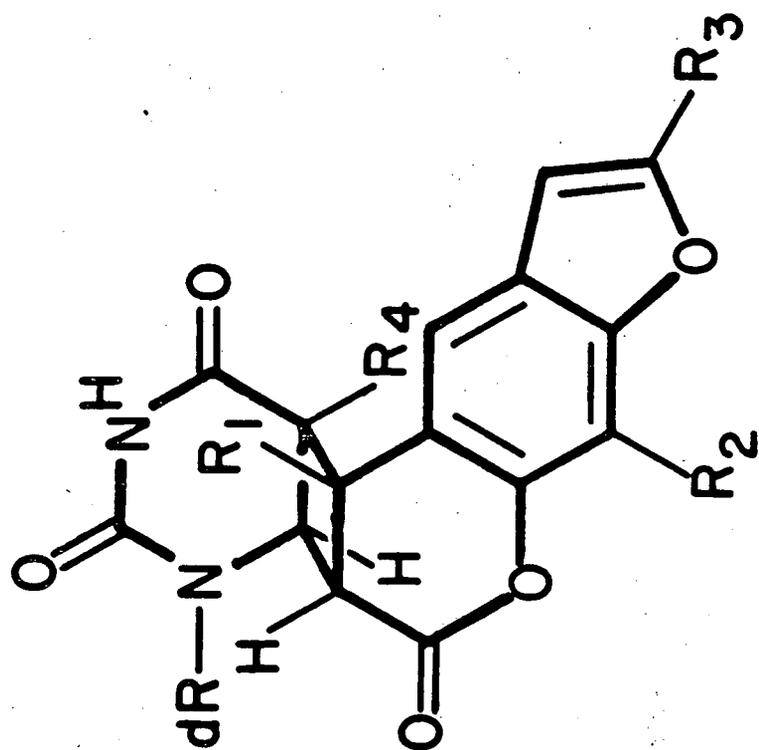
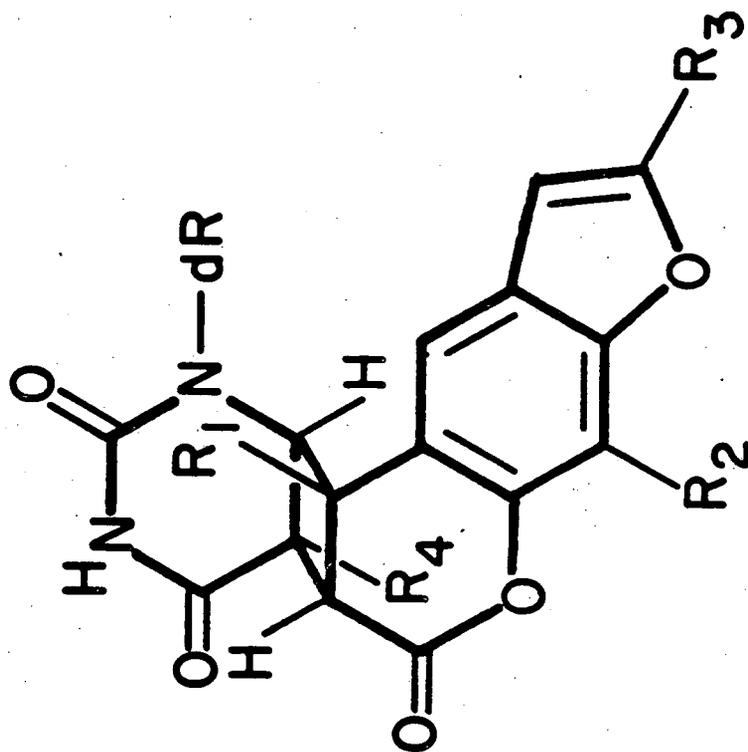
trans-anti-6

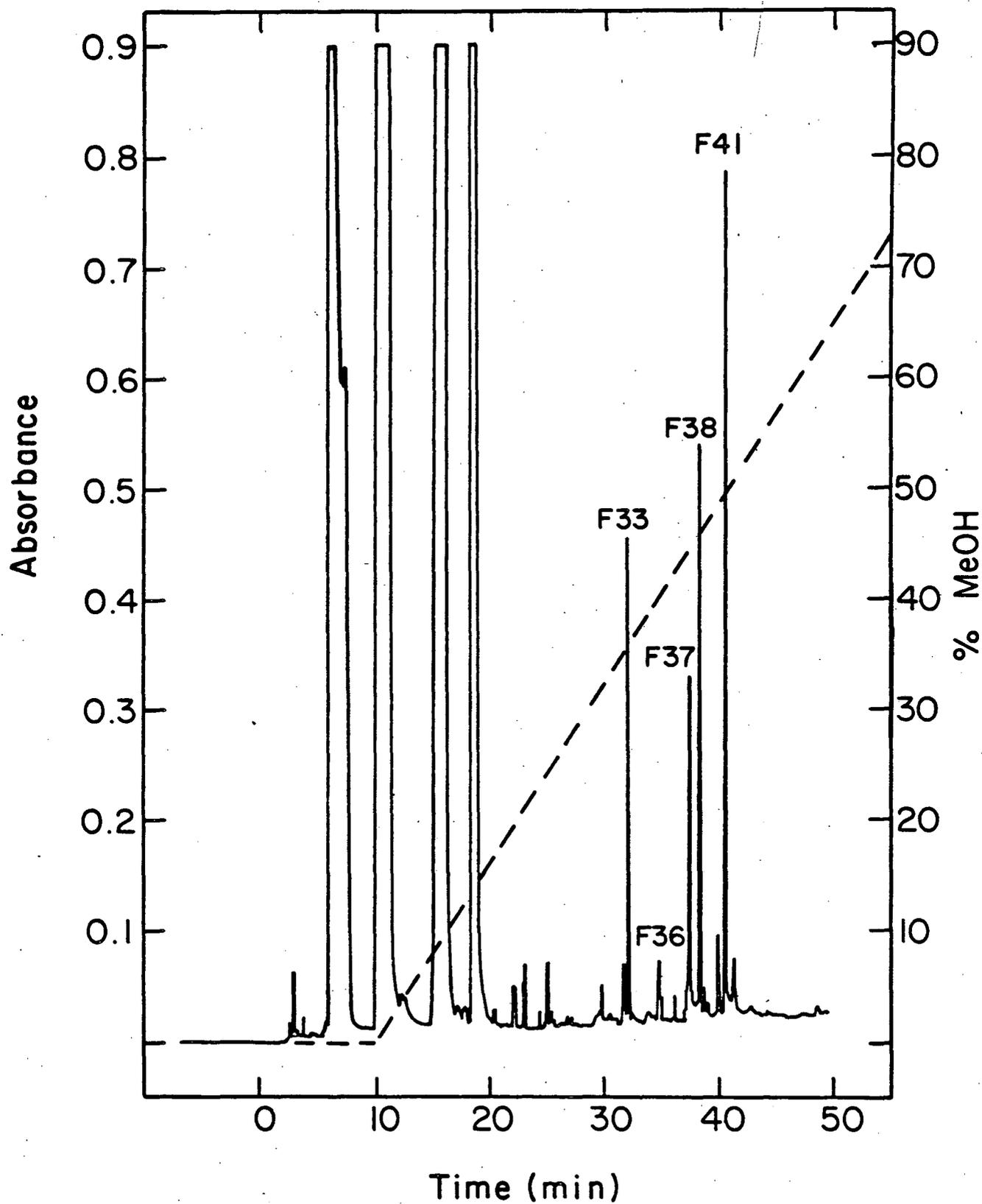


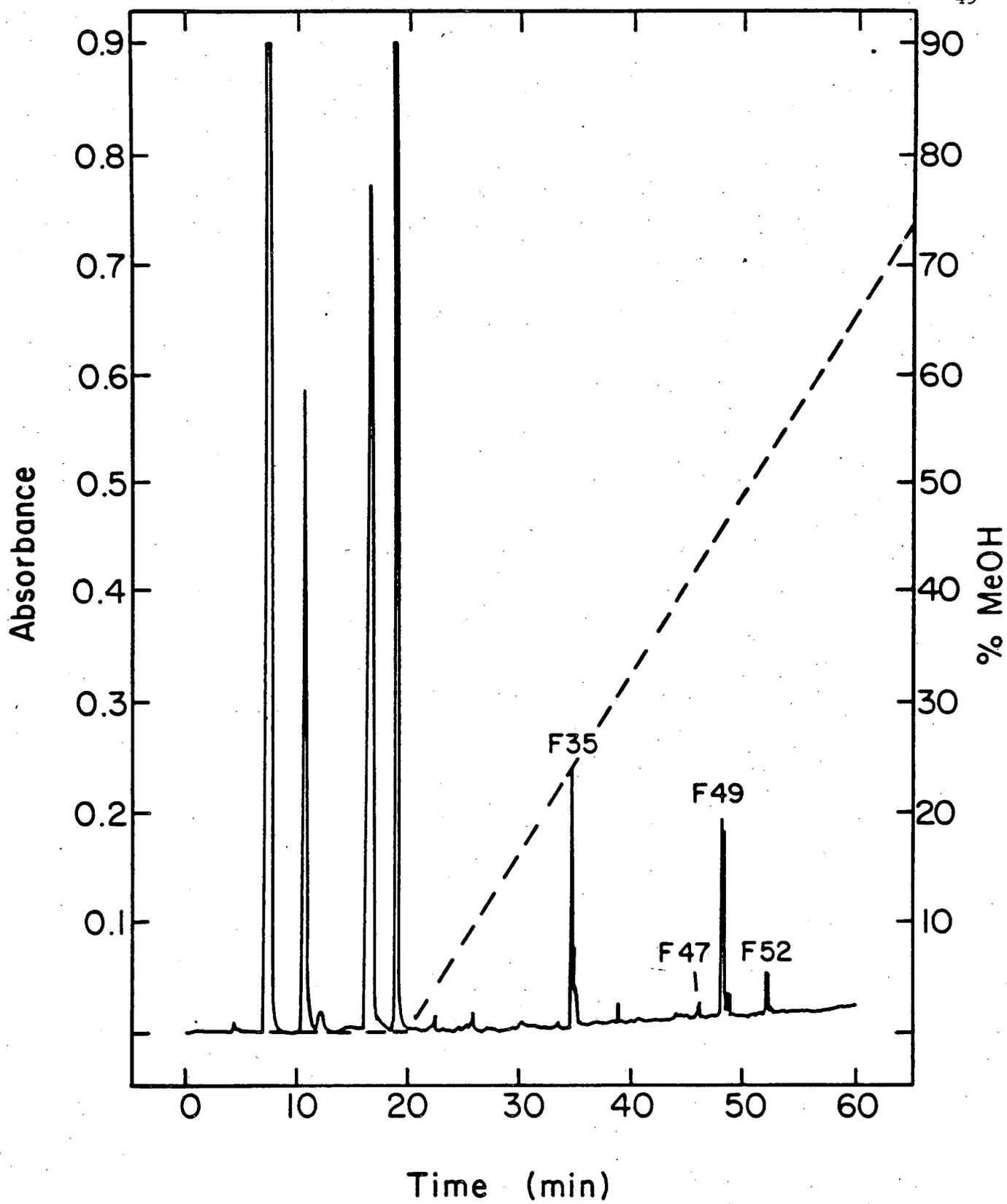
cis-syn-7

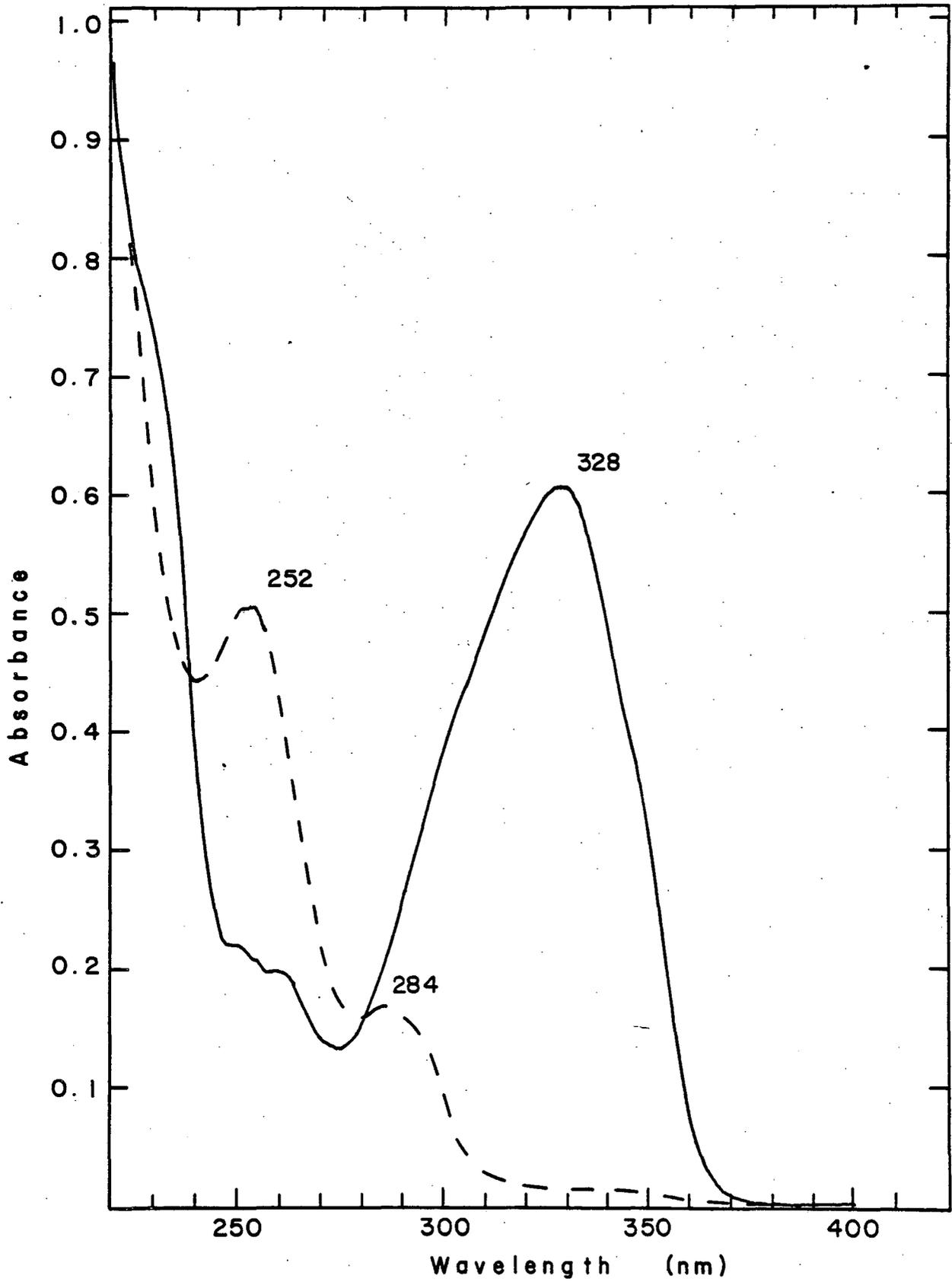


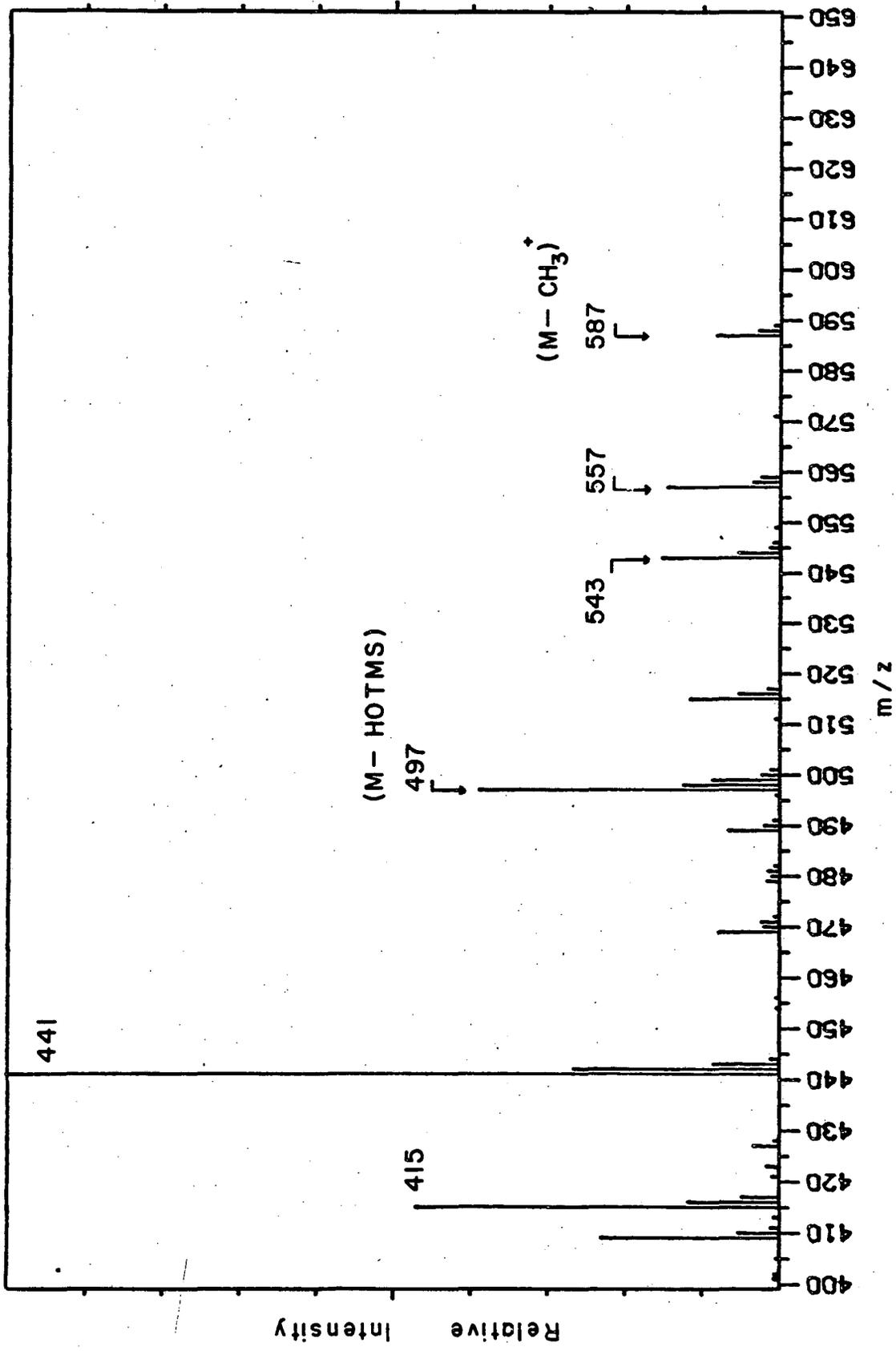
cis-anti-8

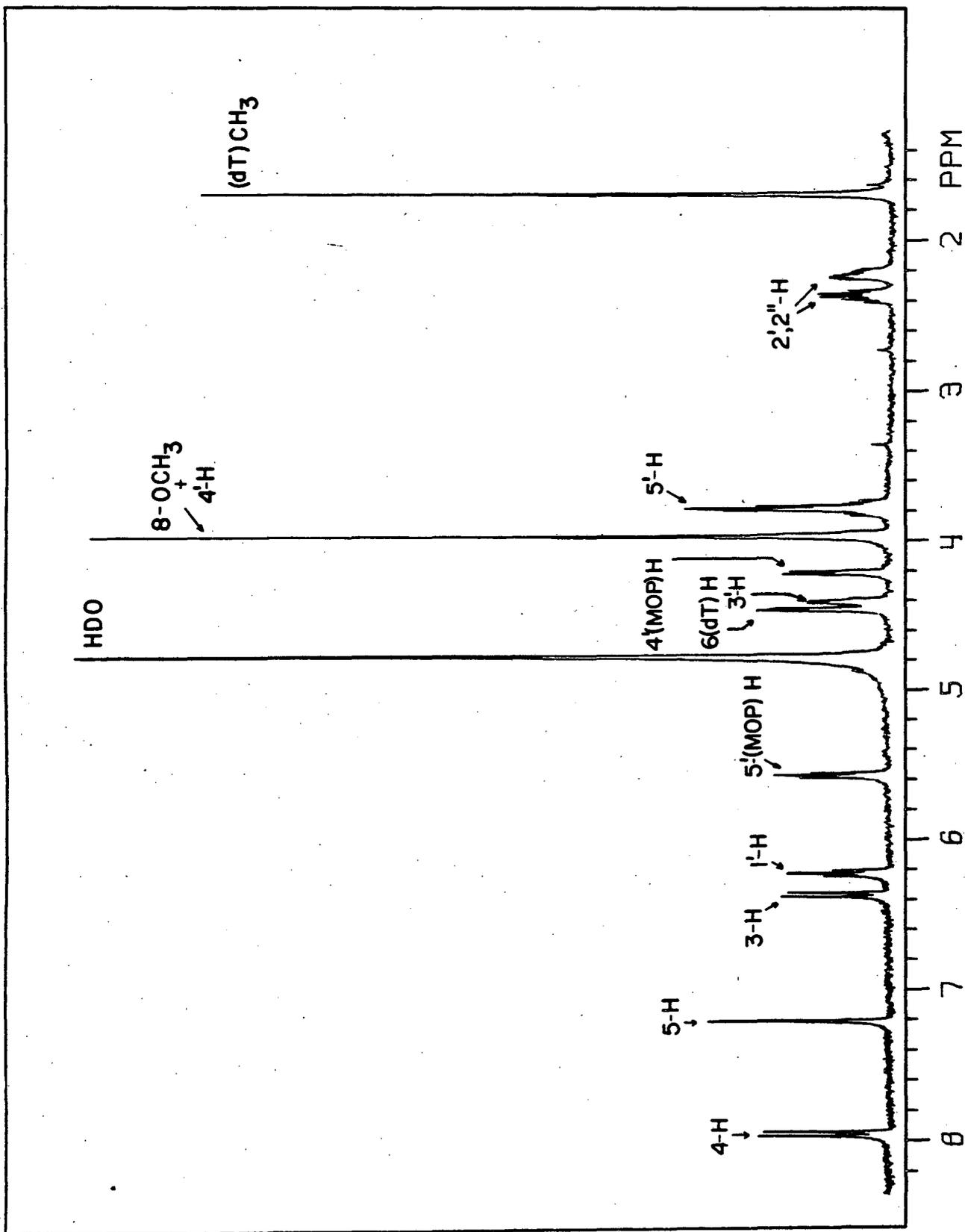
*trans-syn-9**trans-anti-10*

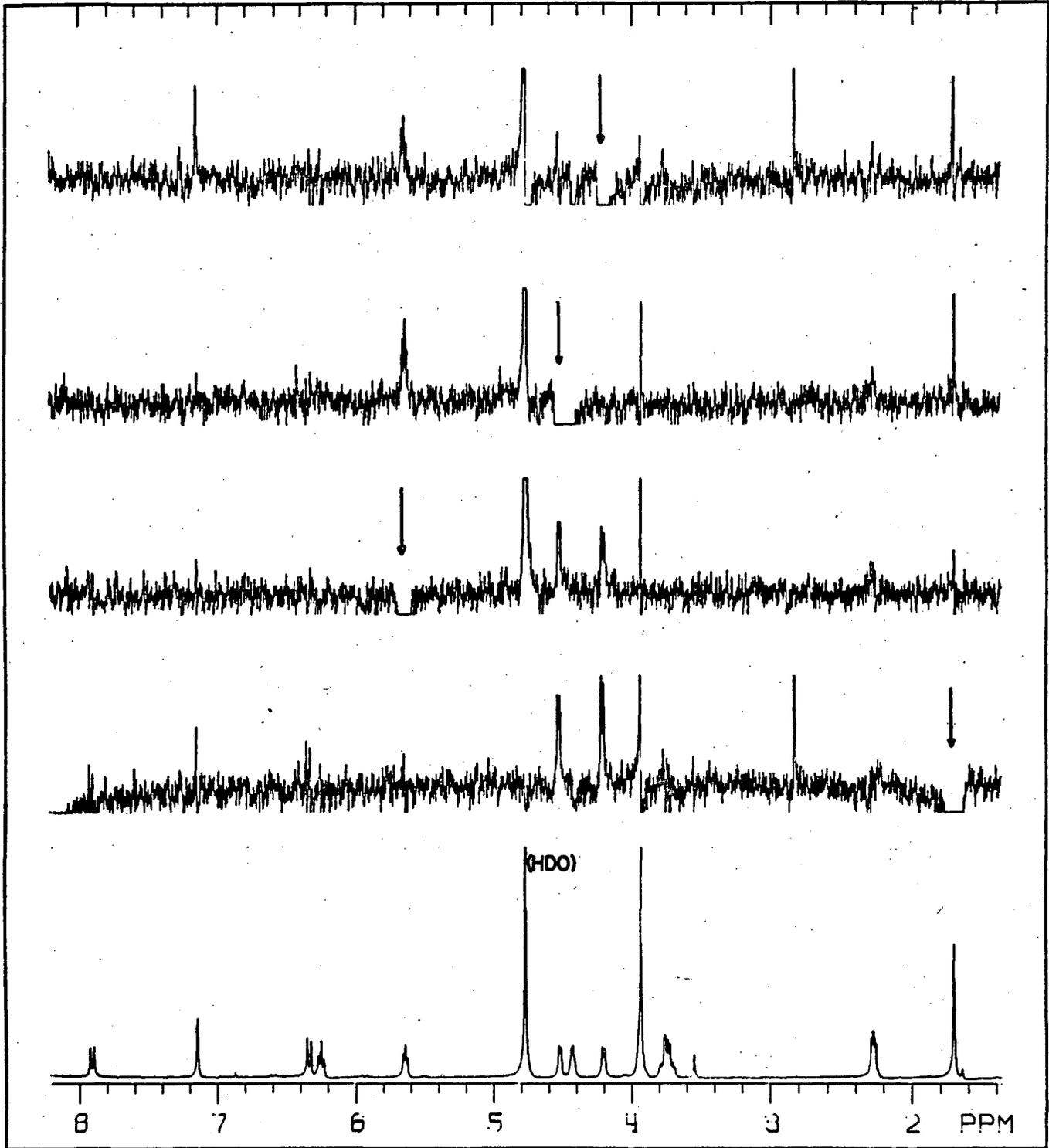


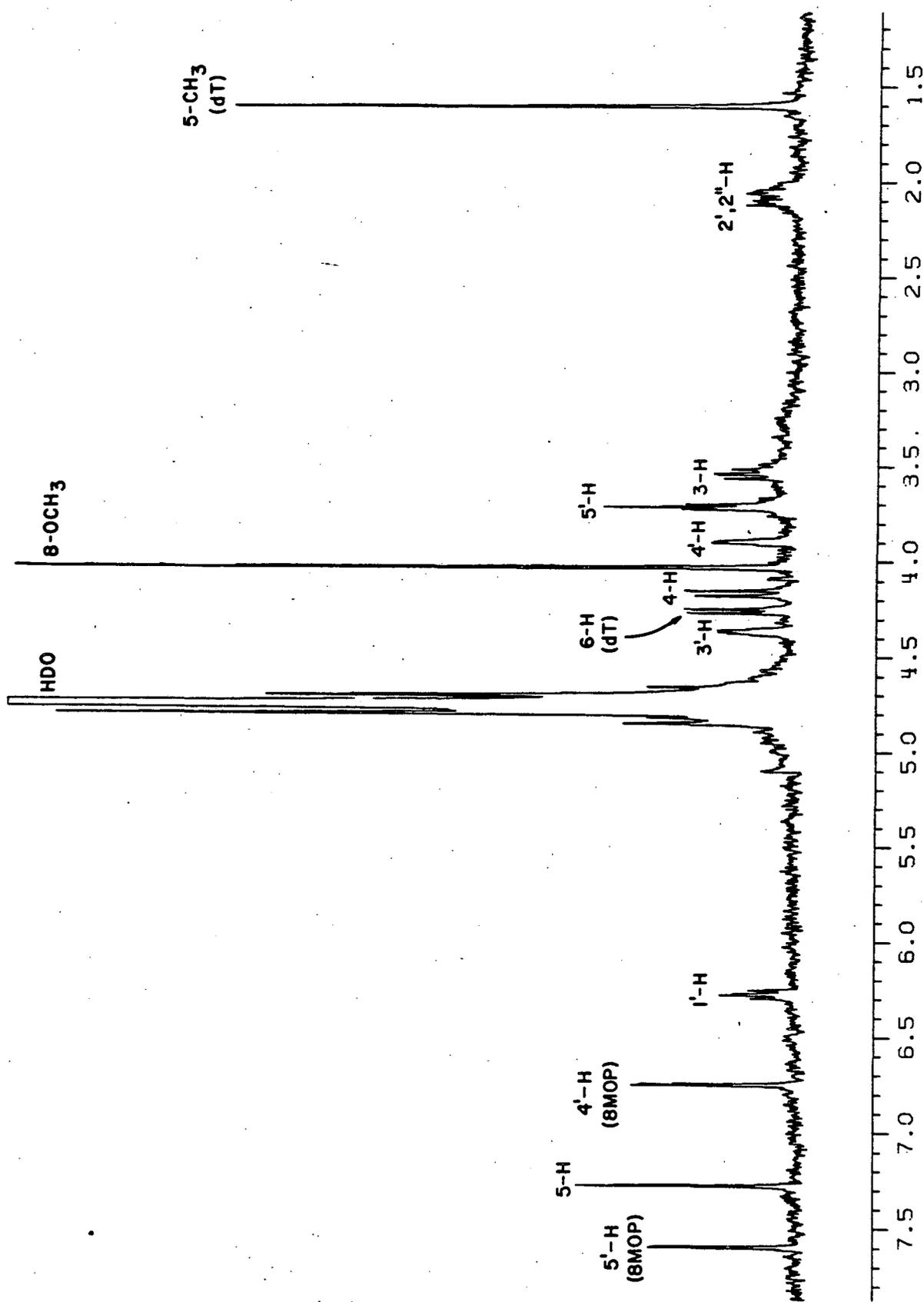


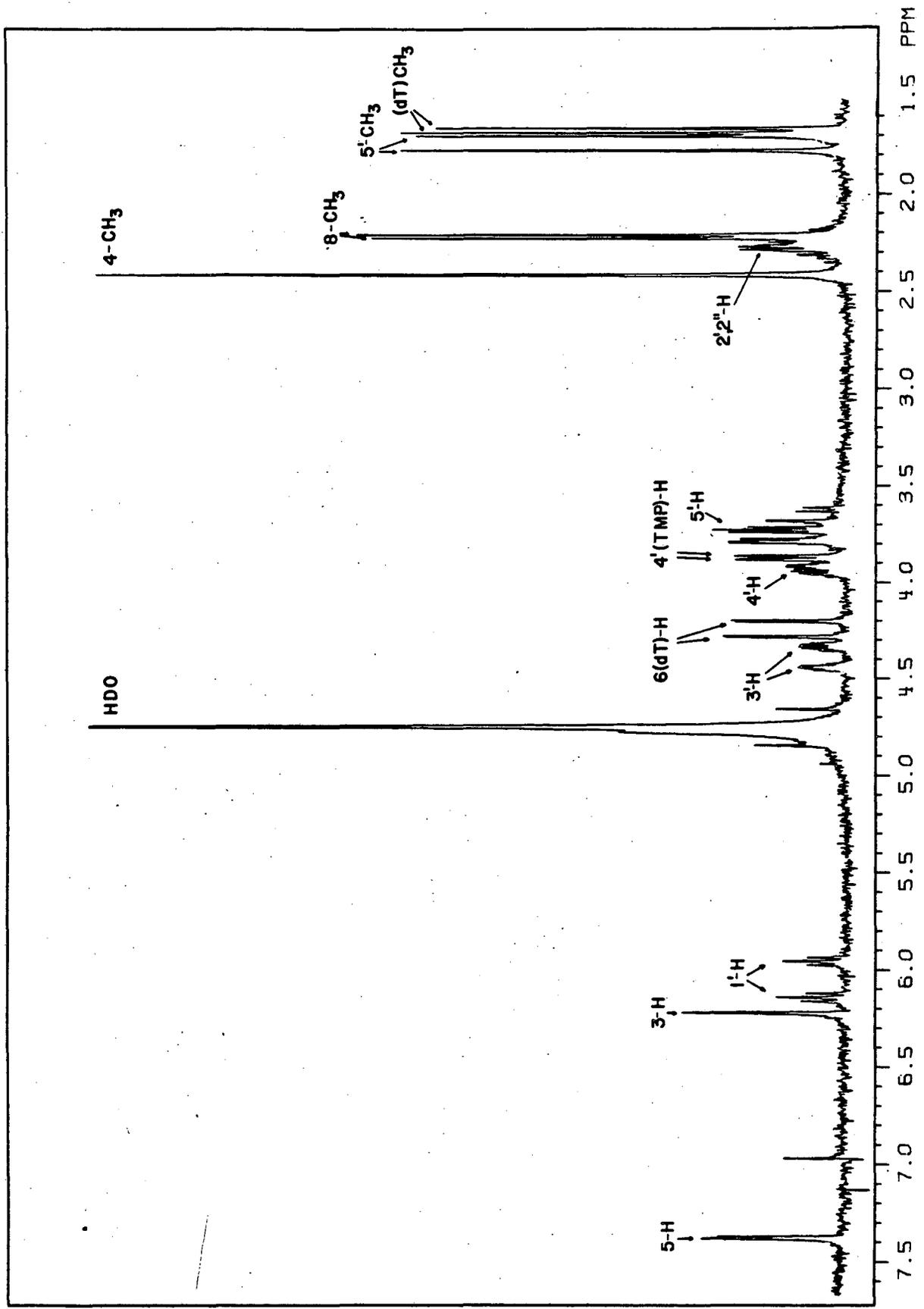


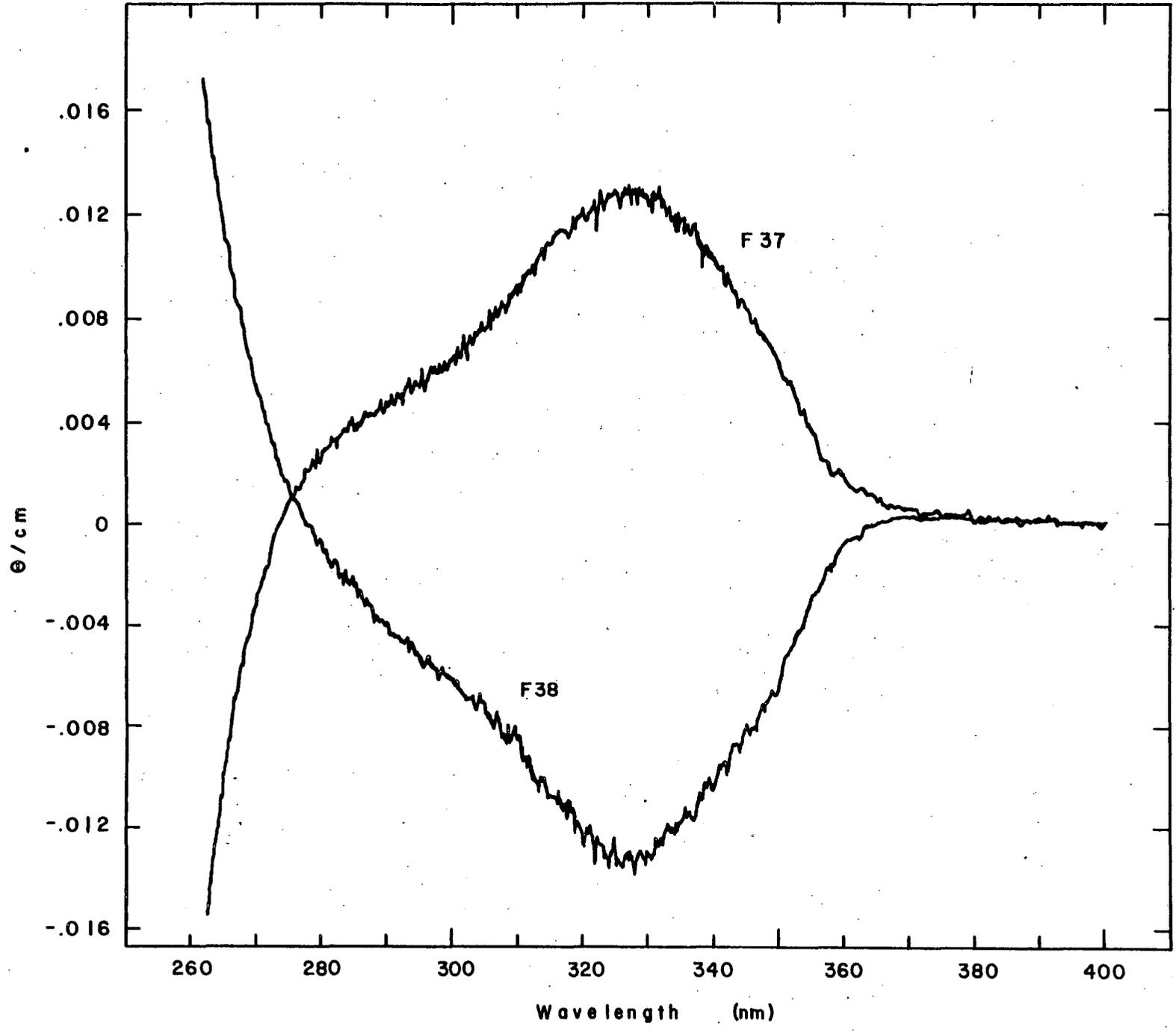












This report was done with support from the Department of Energy. Any conclusions or opinions expressed in this report represent solely those of the author(s) and not necessarily those of The Regents of the University of California, the Lawrence Berkeley Laboratory or the Department of Energy.

Reference to a company or product name does not imply approval or recommendation of the product by the University of California or the U.S. Department of Energy to the exclusion of others that may be suitable.

TECHNICAL INFORMATION DEPARTMENT
LAWRENCE BERKELEY LABORATORY
UNIVERSITY OF CALIFORNIA
BERKELEY, CALIFORNIA 94720